

ENVIRONMENTAL CHEMISTRY

CHEMISTRY

Single Correct Answer Type

- Aerosols present in atmospheric air may be:
 - Positively charged
 - Negatively charged
 - Neutral
 - Combination of all (a), (b) and (c)
- The process which does not evolve CO_2 in air is:
 - Burning
 - Breathing
 - Organic decay
 - Photosynthesis
- High concentration hydrocarbon pollutants in atmospheric air causes:
 - Cancer
 - Silicosis
 - Respiratory diseases (e.g. Asthma)
 - Reduced crop yield
- Ozone layer of stratosphere requires protection from indiscriminate use of:
 - Pesticides
 - Atomic explosions
 - Aerosols and high flying jets
 - Balloons
- Which of the following is the hottest region of the atmosphere?
 - Mesosphere
 - Stratosphere
 - Thermosphere
 - Troposphere
- Lead exhausted in the atmosphere by automobiles using leaded petrol (i.e., tetraethyl lead for improving octane number) is a lethal air pollutant, which causes:
 - Paralysis of muscles and loss of appetite
 - Nervous depression
 - Gastritis and diarrhea
 - All of the above
- Green house effect is accelerated by:
 - Deforestation
 - Rapid industrialization
 - Increased transportation activity
 - All of the above
- Mercury is emitted into air by:
 - Burning coal
 - Burning garbage
 - Coal fire
 - Steam engine
- Chlorofluorocarbon releases which of the following chemical harmful to ozone?
 - Fluorine
 - Chlorine
 - Nitrogen dioxide
 - Sulphur dioxide
- Green house effect is caused by:
 - NO_2
 - CO
 - NO
 - CO_2
- 'Particulate' air pollutants are finely divided solids ($<10^{-6}\text{m}$ size) and liquids. Which of the given is not a 'particulate'?
 - Dust and mists
 - Smoke and fumes
 - Photochemical smog and soot
 - None of the above
- Bhopal gas tragedy of 1984 was caused by:
 - Carbon monoxide
 - Phosgene
 - Methyl cyanate
 - Methyl isocyanate
- Higher concentration of nitrogen dioxide in atmosphere air causes:
 - Cancer
 - Bronchitis
 - Asphyxiation
 - Corrosion

14. London smog is found in:
 a) Summer during day time
 b) Summer during morning time
 c) Winter during morning time
 d) Winter during day time
15. Burning of fossil fuels is the main source of
 a) Nitrogen oxide b) Nitric oxide c) Nitrous oxide d) Sulphur dioxide
16. Phosphate pollution is caused by
 a) Weathering of phosphate rocks only b) Agricultural fertilizers only
 c) Phosphate rocks and sewage d) Sewage and agricultural fertilizers
17. Which compound is mainly responsible for the depletion of ozone layer?
 a) CO₂ b) CH₄ c) CH₃OH d) CF₂Cl₂
18. Carbon monoxide, emitted by automobiles, prevents transport of oxygen in body due to
 a) Combining with oxygen to form carbon dioxide
 b) Destruction of haemoglobin
 c) Preventing reaction between oxygen and haemoglobin
 d) Forming stable compound with haemoglobin
19. Organomercury compounds are
 a) Herbicides b) Fungicides c) Soil conditioners d) Fumigants
20. Carbonaceous particles having size less than 10⁻⁶ m are called:
 a) Gril b) Aggregates c) Aerosols d) Smoke
21. Which one of the following statements regarding photochemical smog is not correct?
 a) Photochemical smog is formed through photochemical reaction involving solar energy
 b) Photochemical smog does not cause irritation in eyes and throat
 c) Carbon monoxide does not play any role in photochemical smog formation
 d) Photochemical smog is an oxidising agent in character
22. The biotic and abiotic components that are affected adversely from harmful substances are called
 a) Target b) Receptor c) Atmosphere d) Both (a) and (b)
23. Which of the following pollutants is not emitted during volcanic eruptions?
 a) SO₂ b) H₂S c) CO d) Hydrocarbons
24. Radioactive pollution is caused by
 a) Solid pollutants b) Liquid pollutants c) Gaseous pollutants d) None of these
25. Harmful chemical present in tobacco is:
 a) Nicotine b) Atropine c) Tannic acid d) Morphine
26. Which of the following is/are the main agents of soil erosion?
 a) Wind and water b) Rocks c) Sand d) None of these
27. Drawback of DDT as pesticides is that
 a) It is less effective than others
 b) It becomes ineffective after some time
 c) It is a nondegradable substance
 d) It is very costly
28. Which of the following is present in maximum amount in acid rain?
 a) HNO₃ b) H₂SO₄ c) HCl d) H₂CO₃
29. Minamata disease is due to pollution of
 a) Organic waste into drinking water b) Oil spill in water
 c) Industrial waste mercury into fishing water d) Arsenic into the atmosphere
30. Carbon monoxide is harmful to human being as it
 a) Is carcinogenic
 b) Is antagonistic to CO₂
 c) Has higher affinity for haemoglobin as compared to oxygen

- d) Is destructive to O_3
31. Lack of oxygen at high altitude produces:
 a) Bends b) Anoxia c) Asthma d) Artificial respire
32. Atmospheric pollutant is
 a) CO_2 b) CO c) O_2 d) N_2
33. Photochemical smog is related to pollution of
 a) Air b) Water c) Soil d) Nostoc
34. Which of the following statement is false?
 a) London smog is oxidising in nature
 b) London smog contains H_2SO_4 droplets
 c) London smog is formed in winter
 d) London smog causes bronchitis
35. Among the following compounds, which one is not responsible for the depletion of ozone layer?
 a) CH_4 b) $CFCl_3$ c) NO d) Cl_2
36. DDT is
 a) Biodegradable pollutant b) Non biodegradable pollutant
 c) Not a pollutant d) An antibiotic
37. Which of the following is not a major constituent of air pollutants?
 a) Oxides of sulphur b) Oxides of nitrogen c) Carbon monoxide d) Hydrogen sulphide
38. Depletion of ozone layer is due to
 a) Oxides of nitrogen b) Oxides of carbon c) Oxides of sulphur d) None of these
39. Modes of controlling pollution in large cities include
 a) Less use of insecticides
 b) Proper disposal of organic wastes, sewage and industrial effluents
 c) Shifting of factories out of the residential area
 d) All of the above
40. DDT and BHC may act as:
 a) Allergens b) Carcinogens c) Asthematic agents d) None of these
41. Ozone hole is maximum over
 a) Europe b) Antarctica c) India d) Africa
42. $CFCl_2$ is responsible for the decomposition of ozone to oxygen. Which of the following reacts with ozone to form oxygen?
 a) Cl_2 b) Cl^- c) F^- d) Cl^*
43. Atmospheric content of CO_2 is
 a) 0.0034% b) 0.034% c) 0.34% d) 3.4%
44. What is DDT among the following?
 a) A fertilizer
 b) Biodegradable pollutant
 c) Non-biodegradable pollutant
 d) Greenhouse gas
45. Which of the following is a living component of atmosphere?
 a) Lithosphere b) Biosphere c) Hydrosphere d) Troposphere
46. Chlorofluorocarbons (CFCs) are widely used in air conditioners, refrigerators etc because of being
 a) Highly reactive b) Flammable c) Non reactive d) All of these are true
47. When rain is accompanied by a thunderstorm, the collected rain water will have a pH value
 a) Uninfluenced by occurrence of thunderstorm
 b) Which depends of the amount of dust in air
 c) Slightly lower than that of rain water without
 d) Slightly higher than that when the thunderstorm is not there
48. For a healthy aquatic life, the amount of dissolved oxygen in a water body must be equal to

- a) 5 ppm b) 4 ppm c) 3 ppm d) 2 ppm
49. Lead is
 a) Air pollutant b) Water and soil pollutant
 c) Radioactive pollutant d) Noise pollutant
50. Negative soil pollution is
 a) Reduction in soil productivity due to erosion and over use
 b) Reduction in soil productivity due to addition of pesticides and industrial wastes
 c) Converting fertile land into barren land by dumping ash, sludge and garbage
 d) None of the above
51. The region which is greatly affected by air pollution is
 a) Troposphere b) Stratosphere c) Mesosphere d) Thermosphere
52. Most hazardous metal pollutant of automobile exhaust is
 a) Mercury b) Lead c) Cadmium d) Copper
53. Which of the following is not a herbicide?
 a) Sodium chlorate b) Sodium arsenite c) Polyphosphate d) Triazines
54. Which of the following is a sink for CO?
 a) Haemoglobin b) Microorganisms present in the soil
 c) Oceans d) Plants
55. Which of the following statements is false?
 a) The main reason for river water pollution is industrial and domestic sewage discharge
 b) Surface water contains a lot of organic matter, mineral nutrients and radioactive materials
 c) Oil spill in sea water causes heavy damage to fishery
 d) Oil slick in a sea water increases DO value
56. Saline soil contains:
 a) High concentration of salt
 b) Lot of moisture
 c) Hard rocks
 d) None of the above
57. Which of the following is not a natural source of air pollution?
 a) Volcanic eruptions and lightening discharges
 b) Biological decay of vegetable matter
 c) Photochemical oxidation of organic matter
 d) None of the above
58. Which of the following is a viable particulate?
 a) Algae b) Smoke c) Mist d) Fumes
59. Taj Mahal is threatened by pollution from
 a) Chlorine b) Sulphur dioxide c) Hydrogen d) Oxygen
60. The instrument used for measuring soil salinity is
 a) Photometer b) Voltmeter c) Conductivitymeter d) Calorimeter
61. Acid rains are produced by
 a) Excess NO₂ and SO₂ from burning fossil fuels
 b) Excess production of NH₃ by industry and coal gas
 c) Excess release of carbon monoxide by incomplete combustion
 d) Excess formation of CO₂ by combustion and animal respiration
62. Growing more trees help to:
 a) Reduce oxygen in the environment
 b) Increase carbon dioxide in the environment
 c) Reduce carbon dioxide only in the environment
 d) Reduce CO₂ and increase O₂ in the environment
63. The ozone layer forms naturally by

- a) The interaction of CFC with oxygen.
 b) The interaction of UV radiation with oxygen.
 c) The interaction of IR radiation with oxygen.
 d) The interaction of oxygen and water vapour.
64. The progressive warming up of the earth surface is mainly due to:
 a) Automobile exhaust
 b) Blanketing effect of carbon dioxide in atmosphere
 c) Reforestation
 d) Thickening of ozone layer
65. Most abundant water pollutant is
 a) Detergents b) Industrial wastes c) Pesticides d) Oil spills
66. Biodegradable pollutant is
 a) Domestic waste b) DDT c) Mercury salt d) Aluminium foil
67. Identify the incorrect statement from the following.
 a) Oxides of nitrogen in the atmosphere can cause the depletion of ozone layer.
 b) Ozone absorbs the intense ultraviolet radiations of the sun.
 c) Depletion of ozone layer is because of its chemical reactions with chlorofluoro alkanes.
 d) Ozone absorbs infrared radiations.
68. Which of the following is the uppermost region of the atmosphere?
 a) Stratosphere b) Troposphere c) Exosphere d) Thermosphere
69. Depletion of ozone layer over Antarctica takes place
 a) In November b) In the months of September and October
 c) In the months of October and November d) In summers
70. Drained sewage has BOD
 a) More than that of water b) Less than that of water
 c) Equal to that of water d) None of these
71. BOD is
 a) Biological oxygen deficit b) Biosphere oxygen demand
 c) Biological oxygen demand d) None of the above
72. Ozone is an important constituent of stratosphere because it
 a) Prevents the formation of smog over large cities
 b) Removes poisonous gases of the atmosphere by reacting with them
 c) Absorbs ultraviolet radiations which is harmful to human life
 d) Destroys bacteria which are harmful to human life
73. Which one is the most toxic?
 a) Carbon b) CO c) CO₂ d) SO₂
74. Which of the following causes water pollution?
 a) Flyash b) Auto exhausts c) Aeroplanes d) Pesticides
75. Identify the wrong statements in the following.
 a) Chlorofluorocarbons are responsible for ozone layer depletion.
 b) Green house effect is responsible for global warming.
 c) Ozone layer does not permit infrared radiation from the Sun to reach the earth.
 d) Acid rain the is mostly because of oxides of nitrogen and sulphur.
76. A secondary pollutant is
 a) CO b) CO₂ c) PAN d) Aerosol
77. Global warming may result in:
 a) Flood
 b) Cyclone
 c) Decrease in forest productivity

- d) All of the above
78. Cyclone collector is used for minimising
 a) Radioactive pollution b) Air pollution c) Noise pollution d) Water pollution
79. Which of the following methods is most effective to remove particulates?
 a) Bag filter method b) Cyclone collector method
 c) Gravity settling chamber d) Electrostatic precipitators
80. Which of the following is responsible for the depletion of the ozone layer in the upper strata of the atmosphere?
 a) Polyhalogens b) Ferrocenes c) Fullerenes d) Freons
81. Green chemistry means such reactions which:
 a) Reduce the use and production of hazardous chemicals
 b) Are related to depletion of ozone layer
 c) Study the reactions in plants
 d) Produce colour during reactions
82. Result of ozone hole is
 a) Green house effect b) Global warming
 c) Acid rain d) UV rays reach the earth
83. Photochemical smog is formed in
 a) Summer during morning time b) Summer during day time
 c) Winter during morning time d) Winter during day time
84. Pick up the correct statement
 a) CO plays a major role in photochemical smog
 b) London smog has an oxidising character whereas Los Angeles smog is reducing in nature
 c) Classical smog is good for health but photochemical smog not
 d) Los Angeles smog forms in day time whereas London smog forms in early morning hours
85. Pneumoconiosis is caused by inhalation of
 a) Coal dust b) Silica dust c) Cotton fibre dust d) Asbestos dust
86. The water pollutants mainly responsible for the eutrophication are
 a) Cd, Pb and Hg present in industrial waste.
 b) Heavy metals present in mining waste.
 c) Detergents and fertilizers containing phosphate anion.
 d) Polychlorinated biphenyls.
87. Which of the following is not a green house gas?
 a) CO₂ b) Water vapour c) CH₄ d) O₂
88. What does BOD₅ represent?
 a) Biological oxygen depletion in five days
 b) Dissolved oxygen left after five days
 c) Dissolved oxygen consumed in five days
 d) Micro-organisms killed by ozone in sewage treatment plants in five hours
89. Pick out the correct statement?
 a) CO which is major pollutant resulting from the combustion of fuels in automobiles plays a major role in photochemical smog
 b) Classical smog has an oxidising character while the photochemical smog is reducing in character
 c) Photochemical smog occurs in day time whereas the classical smog occurs in early morning hours
 d) During formation of smog the level of ozone in the atmosphere goes down
90. Gas released during Bhopal tragedy was
 a) Methyl isocyanate b) Potassium isothiocyanate

- a) Eye disease b) Arthritis c) Kidney damage d) Hair falling
111. Green chemistry involves
- a) Production of chemicals of our daily use from green house gases
 b) Such chemical processes in which green plants are used
 c) Those reactions which are of biological origin
 d) Use of non toxic reagents and solvents to produce environment friendly products
112. Which of the following pollutants is main product of automobile exhaust?
- a) CO b) CO₂ c) NO d) Hydrocarbons
113. Eutrophication causes reduction in
- a) Nutrients b) Dissolved salts c) Dissolved oxygen d) All of these
114. Select the incorrect statement.
- a) Water is considered pure if it has BOD less than 5 ppm
 b) In COD determination, the pollutants resistant to microbial oxidation are not oxidised by oxidising agent like K₂Cr₂O₇
 c) The lower the concentration of DO, the more polluted is the water sample
 d) The tolerable limit of lead in drinking water is 50 ppm
115. Man dies in the atmosphere of CO, because it:
- a) Dries up the blood
 b) Combines with O₂ present in the body
 c) Reduces the organic matter of tissues
 d) Combines with the haemoglobin of blood, thereby making the later incapable of absorbing O₂
116. Methane gas producing field is
- a) Wheat field b) Paddy field c) Cotton field d) Groundnut field
117. The aromatic compounds present as particulates are
- a) Benzene b) Toluene
 c) Nitrobenzene d) Polycyclic aromatic hydrocarbons
118. Which of the following is secondary air pollutant?
- a) Photochemical smog b) NO₂ c) Dust particles d) SO₂
119. Which among the given is not a natural source of air pollution?
- a) Automobile exhausts b) Vegetation decay c) Forest fire d) Volcanic eruptions
120. The greatest affinity for haemoglobin is shown by
- a) NO b) CO c) O₂ d) CO₂
121. Identify the wrong statement in the following
- a) Chlorofluorocarbons are responsible for ozone layer depletion.
 b) Acid rains is mostly because of oxides of nitrogen and sulphur
 c) Green house effect is responsible for global warming
 d) Ozone layer does not permit infrared radiation from the sun to each the earth
122. The principal gas evolved from sludge digestion tank is:
- a) CO b) CO₂ c) CH₄ d) N₂
123. Pollution is
- a) Removal of top soil
 b) Release of toxic/undesirable materials in environment
 c) Conservation of energy
 d) All of the above
124. PAN stands for
- a) CH₂O
 b) CH₂ = CH – CHO
 c) CH₃CH₂O – N = O
 CH₃ – C – OONO₂
 d) $\begin{array}{c} || \\ \text{O} \end{array}$

125. Which of the following gases present in air protects life on the earth from the harmful effects of UV rays from the sun?
 a) Carbon dioxide b) Nitrogen c) Oxygen d) Ozone
126. Which of the following region is the coldest?
 a) Stratosphere b) Troposphere c) Mesosphere d) Thermosphere
127. Rain containing dissolved air pollutants like oxides of S, N and C are referred to as;
 a) Artificial rain b) Acid rain c) Hails d) None of these
128. Which is not an example of gaseous air pollutant?
 a) Oxides of halogens b) Oxides of sulphur c) Oxides of carbon d) Oxides of nitrogen
129. Which of the following is not a part of green chemistry?
 a) Photochemistry b) Sonochemistry c) Nuclear chemistry d) Biochemistry
130. Which of the following statements about control of particulate pollution is false?
 a) Gravity settling chamber removes larger particles from the air.
 b) Cyclone collector removes fine particles in the diameter range 5-20 microns.
 c) Wet scrubbers are used to wash away all types of particulates.
 d) In electrostatic precipitator, the particulates are made to acquire positive charge which are then attracted by the negative electrode and removed.
131. The point of temperature inversion between troposphere and ionosphere is called
 a) Stratopause b) Mesopause c) Tropopause d) Ionopause
132. Which of the following is not regarded as a pollutant?
 a) NO_2 b) CO_2 c) O_3 d) Hydrocarbons
133. Which of the following is not a pollutant?
 a) Carbon monoxide b) Nitrogen peroxide c) Carbon dioxide d) Sulphur dioxide
134. The contribution of which of the following pollutants is least for causing atmospheric pollution?
 a) Nitrogen oxides b) Sulphur oxides c) Hydrocarbon d) Particulates
135. Sulphur dioxide present in the industrial chimney exhaust causes:
 a) Respiratory and lung disease
 b) Reduction in plant's productivity owing to acid rain
 c) Corrosion of building materials
 d) All of the above
136. Which of the following process is involved in the biochemical treatment of sewage effluents?
 a) Oxidation b) Reduction c) Dehydration d) Fermentation
137. The smog is essentially caused by the presence of
 a) O_2 and O_3 b) O_2 and N_2
 c) Oxides of sulphur and nitrogen d) O_3 and N_2
138. Which of the following is responsible for peeling of ozone umbrella?
 a) PAN b) Coal burning c) CFCs d) CO_2
139. Most efficient and suitable dust removal equipment for removal of flyash from flue gas in a thermal power plant is:
 a) Gravity setting chamber
 b) Cyclone separator
 c) Electronic precipitator
 d) Bag filter
140. Main pollutants released from iron and steel industry are:
 a) CO , CO_2 and SO_2 b) NO , SO_3 and H_2S c) CO_2 , H_2S and NO_2 d) CO_2 , NO_2 and SO_3
141. Identify the correct decreasing order of the following with respect to altitude from atmosphere.
 I. Troposphere
 II. Mesosphere
 III. Thermosphere
 a) II, III, I b) III, II, I c) I, II, III d) I, III, II

142. Which is not a green house gas?
 a) CO₂ b) CH₄ c) N₂O d) Chlorofluorocarbons
143. Which of the following metals is not a pollutant?
 a) Mercury b) Arsenic c) Lead d) Aluminium
144. Smog is:
 a) Nothing but black smoke
 b) A combination of smoke and fog
 c) A liquid particle resulting from vapour condensation
 d) A solid particle, e.g., flyash
145. Ozone hole refers to
 a) Hole in ozone layers
 b) Reduction in thickness of ozone layer in stratosphere
 c) Reduction in thickness of ozone in troposphere
 d) Increase concentration of ozone
146. Which of the following is a biodegradable pollutant?
 a) Plastic b) Sewage c) Asbestos d) Mercury
147. Which of the following is a man-made source of air pollution?
 a) Automobile exhaust
 b) Forest fire
 c) Bacterial action in soil and swamp areas
 d) All of the above
148. Which of the following is secondary pollutant?
 a) CO₂ b) N₂O c) PAN d) SO₂
149. The basic component of the smog is
 a) PAN b) PBN c) NO₂ d) All of these
150. Highly toxic gas which causes headache, visual difficulty, paralysis and even death is:
 a) CO₂ b) O₃ c) CO d) All of these
151. The main source by which Taj Mahal may be destroyed is
 a) Flood in Yamuna b) Flue gases from Mathura refinery
 c) Excessive use of natural gas d) All of the above
152. Which of the following gases cause pollution when present in the exhaust fumes of vehicles?
 a) CO₂ b) CO c) Water vapours d) C₂H₆
153. Main source of lead pollution is from:
 a) Sewage b) Leaded gasoline c) Tobacco d) Insecticides
154. Main pollutants released from petroleum refineries are
 a) CO, SO₂ and H₂S b) CO₂, NO and SO₃ c) CO₂, H₂S and NO₂ d) SO₃, NO₂ and CO₂
155. Particulate matter with size less than micron that remain suspended in air indefinitely and transported by wind currents are called:
 a) Fumes b) Mist c) Aerosols d) Soot
156. The depletion of ozone layer in the stratosphere would lead to:
 a) Increased human cataracts and skin cancer
 b) Reduction of planktons in ocean water
 c) Depletion of plants and crops
 d) All of the above
157. The following do/does not cause water pollution:
 a) Heavy metals such as Cd, Pb, Hg
 b) Detergents
 c) Polychlorobiphenyls
 d) Freons
158. Aerobic oxidation is caused by:

- a) Aerobic bacterias in presence of excess of oxygen
 b) Anaerobic bacterias in presence of insufficient oxygen
 c) Aerobic bacterias in the absence of oxygen
 d) Both anaerobic and aerobic bacterias in any condition
159. UV radiation from sun causes a reaction that produces
 a) Carbon monoxide b) Sulphur dioxide c) Fluorides d) Ozone
160. Ozone depletion in the stratosphere is mainly caused by:
 a) SO₂ b) NO₂ c) NO d) chlorofluorocarbons
161. The oxygen present today in atmosphere:
 a) Is a plant product
 b) Came from ozone
 c) Was present in the beginning
 d) Produced by carbon dioxide
162. Ozone in stratosphere is depleted by
 a) CF₂Cl₂ b) C₇F₁₆ c) C₆H₆Cl₆ d) C₆F₆
163. As it passes into food chain, the concentration of DDT
 a) Remains same b) Decreases c) Increases d) Unpredictable
164. Which of the following statements about polar stratospheric clouds (PSCs) is not correct?
 a) Type I clouds are formed at about -77°C and contain solid HNO₃ · 3H₂O
 b) Type II clouds are formed at about -85°C and contain some ice
 c) A tight whirlpool of wind called polar vortex is formed which surrounds Antarctica
 d) PSCs do not react with chlorine nitrate and HCl
165. In Antarctica ozone depletion is due to the formation of following compound
 a) acrolein b) Peroxyacetyl nitrate
 c) SO₂ and SO₃ d) Chlorine nitrate
166. In the upper layer of the atmosphere, ozone is formed by the:
 a) Action of UV rays on oxygen
 b) Combination of oxygen molecules
 c) Action of electric discharge of oxygen molecules
 d) Effect of high pressures on oxygen
167. Chief source of water and soil pollution is
 a) Mining b) Thermal power plant
 c) Agro-industry d) All of these
168. Air pollution from the leakage of methyl isocyanate gas from the Union carbide factory in Bhopal caused a major tragedy on:
 a) Dec. 2, 1984 b) Dec. 15, 1983 c) Dec. 10, 1982 d) Dec. 4, 1988
169. The term acid rain was coined by
 a) Robert Boyle b) Robert Augus c) Alfred Nobel d) Havoisier
170. The lowest layer of earth's atmosphere is
 a) Troposphere b) Mesosphere c) Stratosphere d) Ionsphere
171.plant emits large amount of SO₂ as an air pollutant.
 a) Nitric acid b) Sulphuric acid c) Chloroalkali d) Iron and steel
172. 'Los Angeles' smog is
 a) Sulphurous smog b) Photochemical smog c) Industrial smog d) All of these
173. Solar ultraviolet radiation is absorbed by:
 a) Exosphere b) Ionosphere c) Ozonosphere d) Stratosphere
174. The concentration of salts in soil is increased by
 a) Canal water b) Excessive use of pesticides
 c) Excessive use of fungicides d) All of the above
175. In which part of the atmosphere, ozone layer is present?

- a) Stratosphere b) Troposphere c) Mesosphere d) Thermosphere
176. Persons working in cement plants and limestone quarries are more prone to disease like:
a) Cancer b) Asthma c) Silicosis d) Pneumoconiosis
177. Many people died in London in 1952 as a result of air pollution producing.....
a) Mist b) Fog c) Smog d) Sleet
178. The total amount of oxygen (in ppm), consumed by a pollutant in a water sample is termed as
a) Dissolved oxygen (DO) b) Biochemical oxygen demand (BOD)
c) Chemical oxygen demand (COD) d) None of the above
179. Ozone layer is present in
a) Troposphere b) Stratosphere c) Mesosphere d) Exosphere
180. Addition of phosphate fertilizers into water leads to
a) Increased growth of decomposers b) Reduced algal growth
c) Increased algal growth d) Eutrophication
181. The smog is essentially caused by the presence of
a) O_2 and O_3 b) O_2 and N_2
c) Oxides of sulphur and nitrogen d) O_3 and N_2
182. Temperature of troposphere decreases with altitude. This is because of
a) High pressure of air b) Gases present in air c) Lower density of air d) All of these
183. Which of the following is a primary pollutant?
a) CO b) PAN c) Aldehydes d) H_2SO_4
184. The gas that is not considered as a 'green house gas' is
a) CO_2 b) CH_4 c) O_2 d) O_3
185. Presence of high concentration of ozone and smog in atmospheric air causes:
a) Embrittlement and decrease of folding resistance of paper
b) Cracking of rubber products
c) Fading of dye on textiles
d) Damage of electrical insulator on high tension power line.
186. Which of the following statements is not false?
a) SO_2 does not affect larynx
b) SO_2 is more harmful air pollutant than SO_3
c) NO_2 is more toxic to living tissues than NO
d) NO_x do not play any role in photochemical smog
187. Photochemical smog always contains
a) O_3 b) CO c) CO_2 d) CH_4
188. Which of the following is not a soil pollutant?
a) Polythene bags b) Pesticides
c) Detergents d) Nitrate and phosphate fertilizers
189. Among the following statements which one is incorrect?
a) Suspended particulate matter (SPM) is an important pollutant released by diesel vehicles.
b) Soot particles (size < 5 μ) cause fibrosis of the lung lining
c) H_2SO_4 particulates have size of 500-1000 nm.
d) Photochemical smog is formed by oxides of sulphur, smoke and dust particles.
190. Sewage water is purified by
a) Microorganism b) Light c) Fishes d) Aquatic plants
191. The number of ozone molecules converted into oxygen by one molecule of chlorofluorocarbon compound is
a) One hundred (approximately) b) Ten thousand (approximately)
c) Hundred thousand (approximately) d) Only one
192. Use of chlorofluoro carbons is not encouraged because
a) They are harmful to the eyes of people that use it.

- b) They damage the refrigerators and air conditioners.
 c) They eat away the ozone in the atmosphere.
 d) They destroy the oxygen layer.
193. An important product in the ozone depletion by chlorofluorocarbons is
 a) Cl_2 b) OCl c) OF_2 d) O_2F_2
194. The size of particulates of H_2SO_4 fog lies in the range:
 a) 5–100 nm b) 100–500 nm c) 500–1000 nm d) 1000–10,000 nm
195. Which one of the following can cause depletion of ozone?
 a) H_2S b) NO c) Smoke d) Aerosols
196. Oxides of sulphur and nitrogen are important pollutants of
 a) Water b) Air c) Soil d) All of these
197. Measurement of rate of oxygen utilisation by a unit volume of water over a period of time is to measure
 a) Fermentation b) Biogas generation
 c) Biosynthetic pathway d) Biological oxygen demand
198. The most abundant pollutant is
 a) Ethane b) Methane c) Propane d) Butane
199. Which of the following types of pollution is caused by invisible pollutants?
 a) Thermal pollution b) Noise pollution c) Radioactive pollution d) All of these
200. Which of the following statements is not true?
 a) Ammonia acts as sink for NO_x
 b) Limestone acts as sink for SO_x
 c) The average residence time of NO is one month
 d) SO_x can be removed from flue gases by passing through a solution of citrate ions
201. The photochemical smog can be suppressed by
 a) Nitrogen oxides b) Hydrocarbons
 c) Radical traps d) Formaldehyde
202. Proper management of disposal of household and industrial wastes can be done by
 a) Recycling the waste material to give useful products again
 b) Burning and incineration of combustible waste
 c) Sewage treatment
 d) All of the above
203. Ultraviolet light causes
 a) Formation of pyrimidines
 b) Sticky metaphases
 c) Photodynamic action
 d) Destruction of hydrogen bonds between complementary DNA strands
204. Classical smog occurs in places of
 a) Excess SO_2 b) Low temperature c) High temperature d) Excess NH_3
205. Spraying of DDT produces pollution of the type
 a) Air and water b) Air c) Air, water and soil d) Air and soil
206. Which of the following produces another air pollutant by reacting with oxides of nitrogen in presence of sunlight?
 a) HCl b) SO_2 c) O_3 d) HCN gas
207. White lung cancer is caused by
 a) Asbestos b) Silica c) Paper d) Textiles
208. Water is often treated with chlorine to
 a) Increases oxygen content b) Kill germs
 c) Remove suspended particles d) Remove hardness
209. When rain is accompanied by a thunderstorm, the collected rain water will have a pH value
 a) Slightly lower than that of rain water without thunderstorm.

- b) Slightly higher than that when the thunderstorm is not there.
 - c) Uninfluenced by occurrence of thunderstorm.
 - d) Which depends on the amount of dust in air.
210. Which of the following acts as a sink for chlorine free radicals?
- a) Nitrogen dioxide b) Methane c) Carbon dioxide d) Both (a) and (b)
211. A 'body' which allows the short wavelength incoming solar radiation to enter in but does not allow long wave outgoing infra-red radiation to escape out is called:
- a) Global warming b) Green house c) Atmospheric effect d) Ionosphere

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ENVIRONMENTAL CHEMISTRY

CHEMISTRY

: ANSWER KEY :

1)	d	2)	d	3)	a	4)	c	109)	a	110)	c	111)	d	112)	c
5)	c	6)	d	7)	d	8)	b	113)	c	114)	b	115)	d	116)	b
9)	b	10)	d	11)	a	12)	d	117)	d	118)	d	119)	c	120)	a
13)	b	14)	c	15)	d	16)	d	121)	d	122)	c	123)	b	124)	d
17)	d	18)	d	19)	b	20)	d	125)	d	126)	c	127)	b	128)	a
21)	b	22)	d	23)	d	24)	d	129)	c	130)	d	131)	c	132)	b
25)	a	26)	a	27)	c	28)	b	133)	b	134)	a	135)	d	136)	a
29)	c	30)	c	31)	b	32)	b	137)	c	138)	c	139)	c	140)	a
33)	a	34)	a	35)	a	36)	b	141)	b	142)	c	143)	d	144)	b
37)	d	38)	a	39)	d	40)	b	145)	b	146)	b	147)	a	148)	c
41)	b	42)	d	43)	b	44)	c	149)	d	150)	c	151)	b	152)	b
45)	b	46)	c	47)	c	48)	a	153)	b	154)	a	155)	c	156)	d
49)	b	50)	a	51)	a	52)	b	157)	d	158)	a	159)	d	160)	d
53)	c	54)	b	55)	d	56)	a	161)	a	162)	a	163)	c	164)	d
57)	d	58)	a	59)	b	60)	c	165)	d	166)	a	167)	d	168)	a
61)	a	62)	d	63)	b	64)	b	169)	b	170)	a	171)	b	172)	b
65)	b	66)	a	67)	d	68)	c	173)	d	174)	a	175)	a	176)	c
69)	b	70)	a	71)	c	72)	c	177)	c	178)	c	179)	b	180)	d
73)	b	74)	d	75)	c	76)	c	181)	c	182)	c	183)	a	184)	c
77)	d	78)	b	79)	d	80)	d	185)	b	186)	c	187)	a	188)	c
81)	a	82)	d	83)	b	84)	d	189)	d	190)	a	191)	c	192)	c
85)	a	86)	c	87)	d	88)	c	193)	b	194)	c	195)	d	196)	b
89)	d	90)	a	91)	c	92)	b	197)	d	198)	b	199)	d	200)	c
93)	c	94)	b	95)	a	96)	b	201)	c	202)	d	203)	d	204)	b
97)	a	98)	b	99)	b	100)	d	205)	c	206)	c	207)	d	208)	b
101)	c	102)	c	103)	c	104)	b	209)	a	210)	d	211)	b		
105)	b	106)	d	107)	a	108)	c								

ENVIRONMENTAL CHEMISTRY

CHEMISTRY

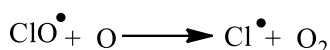
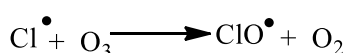
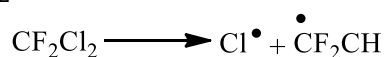
: HINTS AND SOLUTIONS :

4 (c)
Aerosols and high flying jets release nitric oxide into the upper atmosphere which leads to the destruction of ozone layer

10 (d)
Green house effect is caused by CO₂.

12 (d)
Bhopal gas tragedy of 1984 was caused by methyl isocyanate (MIC). This gas was released from a pesticide manufacturing plant union carbide.

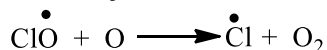
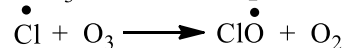
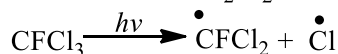
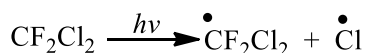
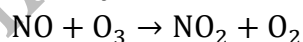
17 (d)
Chlorofluorocarbons provide Cl[•] free radical, which reacts with ozone and converts it into O₂ as



The reaction, once start, continues for a long time. Thus, chlorofluorocarbons (CF₂Cl₂) are responsible for the depletion of ozone layer.

18 (d)
Carbon monoxide is highly toxic to living being because it has an ability to form more stable carboxyhaemoglobin complex with haemoglobin due to which the delivery of oxygen to the organs and tissues is blocked

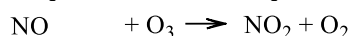
35 (a)
In stratosphere the following reactions takes place which are responsible for the depletion of ozone layer



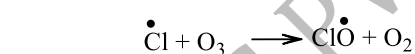
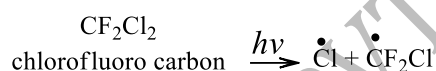
Hence, methane (CH₄) is not responsible for ozone layer depletion.

38 (a)

Oxides of nitrogen and chlorofluorocarbons, when reach into the stratosphere, react with ozone molecules and convert them into oxygen. Thus, these compounds lead to depletion of ozone layer.



oxide of nitrogen



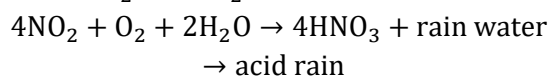
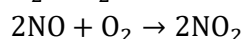
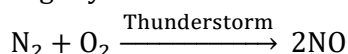
41 (b)
Ozone hole is maximum over Antarctica

42 (d)
CFC (Chlorofluorocarbons) are causing depletion of ozone in the stratosphere. This occurs because ultra violet light also causes CFC's to decompose, producing atomic chlorine. The chlorine atoms react with ozone molecules, resulting in a net removal of O₃ molecules from the stratosphere.

45 (b)
Biosphere is a living component of atmosphere

46 (c)
Chlorofluoro, carbons *ie*, freons are non reactive, non inflammable, non toxic organic molecules, these are widely used in air conditioners, refrigerators

47 (c)
When rain is accompanied by thunderstorm, the N₂ and O₂, present in the atmosphere, combine together to give oxides of nitrogen which dissolve in rain water and give nitric acid (a strong acid). Due to presence of acid, the pH of rain water gets slightly lower



53 (c)
Herbicides are used to kill weeds, eg, sodium chlorate, sodium arsinite, triazines are used as herbicides

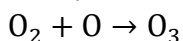
54 (b)

Microorganisms present in the soil act as a sink for carbon monoxide

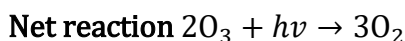
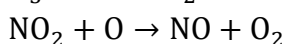
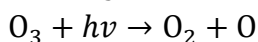
55 (d) Oil slick causes water pollution, thus it decreases DO value (dissolved oxygen value) of sea water

60 (c) Conductivitymeter is used for measuring soil salinity

63 (b) $O_2 \xrightarrow[\text{rays}]{hv} O + O$



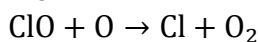
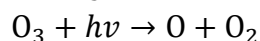
67 (d) (a) $NO + O_3 \rightarrow NO_2 + O_2$



Thus, ozone layer is depleted by oxides of nitrogen.

(b) Ozone layer is a protective layer and absorbs harmful UV rays coming from the sun.

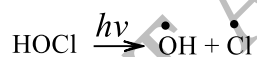
(c) $Cl + O_3 \rightarrow ClO + O_2$



Thus, ozone layer is also depleted by reaction with freons.

(d) is a incorrect statement as

69 (b) During spring season *ie*, in the month of September and October, the sunlight returns to the Antarctica and breaks up the clouds and photolysis HOCl and Cl₂



These $\overset{\bullet}{Cl}$ free radical again reacts with ozone molecules and leads to ozone depletion

73 (b) Carbon monoxide is highly poisonous to living being

75 (c) Ozone layer permits the infrared radiations to pass through but doesn't permit the higher range of ultraviolet radiation to pass through.

78 (b) Cyclone collector is used to remove particulate

particles, thus it minimises air pollution

80 (d) Freons or chlorofluoro carbons are responsible for depletion of the ozone layer in the upper strata of the atmosphere. They are used as propellants, aerosol spray caps, refrigerants, fire fighting reagents etc. They are stable and chemically inert compounds. They absorb UV-radiation and break down liberating free atomic chlorine which causes decomposition of ozone through free radical reaction. This results in the depletion of the ozone layer.

Freons are mainly freon-1 (CFCl₃) and freon-12 (CF₂Cl₂). They form free radical of chlorine in the presence of UV-radiation. Such free radical decomposes O₃ as follows



84 (d) Classical or London type smog is formed by the combination of soot particles with oxides of sulphur while climate is cool and humid. Due to presence of soot and oxides of sulphur, it is reducing in nature.

Photochemical smog or Los Angeles smog is obtained from nitrogen oxides when climate is warm, dry and sunny. Due to presence of O₃ and NO₂ (strong oxidising agents), it is oxidising in nature.

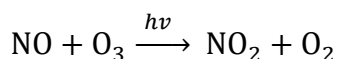
CO does not play any role in the formation of photochemical smog

85 (a) Pneumoconiosis is caused by inhalation of coal dust

86 (c) Detergents and fertilizers contain phosphates as additives. The addition of phosphorous to water, in the form of the phosphate anion (PO₄³⁻), encourages the formation of algae, which reduces the dissolved oxygen concentration of water. This process is known as eutrophication.

88 (c) BOD₅ means, dissolved oxygen consumed in five days

- 89 **(d)**
During the formation of photochemical smog the level of ozone in the atmosphere goes down.

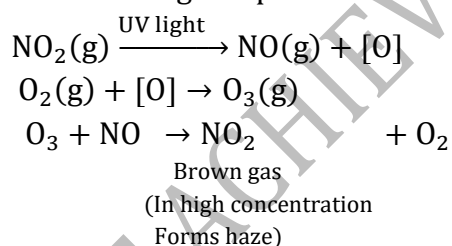


In the atmosphere the organic compounds rapidly react with O_3 , NO_2 to form other noxious photochemical products known as peroxyacyl nitrates (PANs) and acrolein. Photochemical smog occurs in warm, dry and sunny climate, generally during the day time while classical smog occurs in cool humid climate, generally in the early morning hours of winter months.

- 90 **(a)**
Methyl isocyanate (MIC) gas was released during Bhopal tragedy

- 91 **(c)**
Mesosphere and thermosphere are collectively known as ionosphere as in these, gases are present in their ionised form

- 94 **(b)**
Photochemical smog is initiated by the photochemical dissociation of NO_2 and the resulting secondary reactions involving unsaturated hydrocarbons, other organic compounds and free radicals, lead to the formation of organic peroxides and ozone.



Hydrocarbons + O_3 , O_2 , O , NO_2 , NO peroxides, peroxyacetyl nitrate, formaldehyde, ozone aldehyde, acrolein, etc. oxidised hydrocarbons and ozone in the presence of humidity cause photochemical smog, which dissipates at night.

- 97 **(a)**
Nitrogen gas is present in air up to 78% by volume. It does not cause pollution.

- 105 **(b)**
Due to ozone depletion of ozone layer, harmful UV radiations reach the earth surface. These radiations causes skin cancer, sunburn, and also lead to harmful mutation of cell

- 114 **(b)**
In COD determination, the pollutants, which are resistant to microbial oxidation, are also oxidised by strong oxidising agents such as $\text{K}_2\text{Cr}_2\text{O}_7$

- 116 **(b)**
Methane gas producing field is paddy field. It is also known as marsh gas

- 120 **(a)**
Nitrous oxide (NO) has the highest affinity towards haemoglobin. However, due to its larger size, it cannot be inhaled

- 121 **(d)**
Ozone layer is permeable for infrared radiations but it does not allow the harmful UV radiations to reach on the earth

- 126 **(c)**
Mesosphere is the coldest region having -100°C temperature

- 129 **(c)**
Green chemistry involves photochemistry (related to light), sonochemistry (related to sound waves) and biochemistry (related to enzymes) but it does not involve nuclear chemistry

- 130 **(d)**
In electrostatic precipitator, the electrode plate is positively charged. Thus, the particulates acquire negative charge and are attracted by the negative electrode plate

- 132 **(b)**
Carbon dioxide, being limiting factor, when present in small amount (*ie*, 0.033%), has no adverse effect but when its concentration is slightly higher than 0.033%, it has an adverse effect on our climate. Thus, in normal conditions, CO_2 is not regarded as a pollutant

- 141 **(b)**
Atmosphere is divided into four parts

Troposphere 0-10 km

Stratosphere 10-50 km

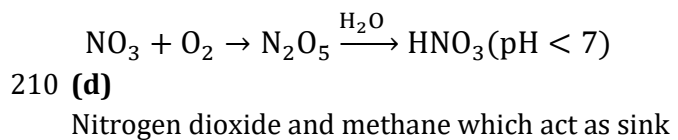
Mesosphere 50-85 km

Thermosphere 85-100 km

- 146 **(b)**
Sewage is a biodegradable pollutant because it easily decompose by microorganism

- 148 **(c)**
Pollutants which are formed by reaction of

- primary pollutants (persist in the environment in the form they are passed into it) are called as secondary pollutants *e. g.*, peroxyacyl nitrates (PAN) are formed through reaction between nitrogen oxides and hydrocarbons in the presence of sunlight.
- 149 **(d)**
Peroxyacetyl nitrate (PAN), peroxybenzoyl nitrate, nitrogen dioxide (NO₂) and hydrogen peroxide (H₂O₂) are the components of smog.
- 162 **(a)**
In stratosphere, chlorofluorocarbons (CF₂Cl₂, CFC₁₂) etc. are responsible for the depletion of ozone layer
- 163 **(c)**
When a pesticide such as DDT passes from lower tropic level to higher level, through food chain, the amount of pesticide per unit weight of organism increases due to accumulation in fat. This process is called biomagnification
- 164 **(d)**
PSCs (polar stratospheric clouds) of type II provide a surface for the conversion of chlorine nitrate (ClONO₂) and HCl into HOCl and Cl₂
- $$\text{ClONO}_2 + \text{H}_2\text{O} \xrightarrow{\text{PSCs}} \text{HOCl} + \text{HNO}_3$$
- $$\text{ClONO}_2 + \text{HCl} \rightarrow \text{Cl}_2 + \text{HNO}_3$$
- 165 **(d)**
ClONO₂ + H₂O → HOCl + HNO₃
ClONO₂ + HCl → Cl₂ + HNO₃
HOCl and Cl₂ can get converted into chlorine radicals and thus, responsible for O₃ depletion.
- 169 **(b)**
The term acid rain was coined by 'Robert August'
- 175 **(a)**
Ozone layer is found in the stratosphere region of atmosphere. It prevents harmful UV radiation from coming to earth.
- 179 **(b)**
In, stratosphere, at about 20 to 40 km, there is a part of relatively high ozone concentration, called ozone layer
- 181 **(c)**
NO, NO₂, SO₂ and SO₃ are responsible for smog (environmental pollution).
- 182 **(c)**
In troposphere, as we move towards the altitude, the density and pressure of air decreases. Due to which temperature also decreases
- 184 **(c)**
Oxygen gas does not absorb I.R. radiation of high wavelengths reflected back by earth, hence it does not cause 'green house effect'.
- 186 **(c)**
SO₂ affects larynx, between SO₂ and SO₃, SO₃ is more harmful air pollutant and between NO₂ and NO, NO₂ is more toxic. Photochemical smog is caused by oxides of nitrogen
- 189 **(d)**
When climate is warm, dry and sunny, the oxides of nitrogen and unsaturated hydrocarbons are converted in the components such as PAN, formaldehyde which form photochemical smog, by the action of sunlight
- 191 **(c)**
One chlorine free radical can convert about one lakh ozone molecules into oxygen
- 192 **(c)**
Chlorofluorocarbon is used in air conditioning and in domestic refrigerators for cooling purposes. Its main drawback in this, it is responsible for ozone depletion.
- 193 **(b)**
We know that,
$$\text{O}_3 + \text{CCl}_2\text{F}_2 \rightarrow 2\text{OCl} + \text{F}_2\text{O}$$
. Thus, in this reaction OCl is produced.
- 200 **(c)**
The average residence time of NO is 4 days
- 201 **(c)**
Photochemical smog can be suppressed by radical traps. When the compounds are sprayed to the atmosphere, they generate free radicals which readily combine with free radical precursors of photochemical smog. Diethyl hydroxylamine has been found to possess smog inhibiting characteristics.
- 203 **(d)**
The hydrogen bonds present between complementary strands of DNA are destructed by UV light
- 209 **(a)**
During thunderstorm, there is formation of NO which changes to NO₂ and ultimately to HNO₃ (acid rain).
$$\text{N}_2 + \text{O}_2 \rightarrow \text{NO} \xrightarrow{\text{O}_2} \text{NO}_2 \xrightarrow{\text{O}_3} \text{NO}_3 + \text{O}_2$$



for chlorine free radicals and prevent much ozone depletion

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ENVIRONMENTAL CHEMISTRY

CHEMISTRY

Assertion - Reasoning Type

This section contain(s) 0 questions numbered 1 to 0. Each question contains STATEMENT 1(Assertion) and STATEMENT 2(Reason). Each question has the 4 choices (a), (b), (c) and (d) out of which **ONLY ONE** is correct.

- a) Statement 1 is True, Statement 2 is True; Statement 2 **is** correct explanation for Statement 1
- b) Statement 1 is True, Statement 2 is True; Statement 2 **is not** correct explanation for Statement 1
- c) Statement 1 is True, Statement 2 is False
- d) Statement 1 is False, Statement 2 is True

1

Statement 1: For green house effect presence of CO₂ is essential.

Statement 2: With increase in concentration of CO₂, green house effect increases.

2

Statement 1: Deforestation, is one main factor contributing to global warming

Statement 2: Besides CO₂ two other gases methane and CFCs one also included under green house gases

3

Statement 1: Presently, the global atmosphere is warming up

Statement 2: The depletion of stratospheric ozone layer has resulted in increases in ultraviolet radiations reaching the earth

4

Statement 1: Water having pH<5.5 is not suitable for drinking purposes.

Statement 2: As the pH of water decreases, the solubility of metal ions increases.

5

Statement 1: Photochemical smog is produced by nitrogen oxides

Statement 2: Vehicular pollution is a major source of nitrogen oxides

6

Statement 1: CO combines with haemoglobin

Statement 2: It has affinity for haemoglobin

7

Statement 1: Suspended particulate matter (SPM) is an important pollutant released by diesel vehicles.

Statement 2: Catalytic converters greatly reduce pollution caused by automobiles.

8

Statement 1: Presently the global atmosphere is warming up.

Statement 2: The depletion of stratospheric ozone layer has resulted in increase in ultraviolet radiations reaching the earth.

9

Statement 1: Carbon monoxide combines with haemoglobin.

Statement 2: CO has more affinity for haemoglobin.

10

Statement 1: Acid rain has a pH less than 5

Statement 2: Oxides of nitrogen and sulphur combine with rain water to produce acidic nature

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ENVIRONMENTAL CHEMISTRY

CHEMISTRY

: ANSWER KEY :

1)	b	2)	b	3)	b	4)	a	9)	a	10)	a
5)	b	6)	a	7)	b	8)	b				

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ENVIRONMENTAL CHEMISTRY

CHEMISTRY

: HINTS AND SOLUTIONS :

- | | | | |
|---|--|----|--|
| 1 | (b)
CO ₂ molecules trap the longer wavelength, infrared radiations emitted by earth and causes green house effect. | | soot flyash, dusts of various types. These are hazardous. They stay in air long enough to travel all over the world and bring with them toxic cancer causing pollutant. |
| 2 | (b)
Both Statement I and II are true but Statement II is not correct explanation | 8 | (b)
Global atmosphere is warming up due to increase in concentration of green house gases. |
| 3 | (b)
Both Statement I and II are true but Statement II is not correct explanation | 9 | (a)
CO has more affinity for hemoglobin than that of O ₂ and from stable compound carboxy haemoglobin and reduces the oxygen carrier capacity of the blood. |
| 4 | (a)
The normal rain water has a pH of 5.6 due to dissolution of CO ₂ in it.

$\text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{CO}_3 \rightleftharpoons \text{H}^+ + \text{HCO}_3^-$ | 10 | (a)
Statement II is correct explanation of Statement I |
| 5 | (b)
Both Statement I and II are true but Statement II is not correct explanation | | |
| 7 | (b)
Suspended particulate matter (SPM) consists of | | |

ENVIRONMENTAL CHEMISTRY

CHEMISTRY

Matrix-Match Type

This section contain(s) 0 question(s). Each question contains Statements given in 2 columns which have to be matched. Statements (A, B, C, D) in **columns I** have to be matched with Statements (p, q, r, s) in **columns II**.

1. Match the list I and II and pick the correct matching from the codes given below

Column-I

- (A) Polycyclic aromatic hydrocarbons
- (B) Dioxins
- (C) IR active molecules
- (D) Peroxy acetyl nitrate

Column- II

- (p) Global warming
- (q) Photochemical smog
- (r) Carcinogens
- (s) Waste incineration

CODES :

- | | A | B | C | D |
|----|---|---|---|---|
| a) | c | d | a | b |
| b) | d | c | b | a |
| c) | c | d | b | a |
| d) | a | b | c | d |

ENVIRONMENTAL CHEMISTRY

CHEMISTRY

: ANSWER KEY :

1) a

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ENVIRONMENTAL CHEMISTRY

CHEMISTRY

: HINTS AND SOLUTIONS :

- 1 (a)
Polycyclic aromatic hydrocarbons are carcinogens, *ie*, cancer producing dioxins are waste incineration, IR active molecules such as CO₂ are related with global warming. PAN (peroxy acetyl nitrate) forms photochemical smog

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