## SECTION - A

1. If the coefficients of $x$ and $x^{2}$ in $(1+x)^{p}(1-x)^{q}$ are 4 and -5 respectively, then $2 p+3 q$ is equal to
(1) 60
(2) 63
(3) 66
(4) 69

Sol. (2)
$(1+\mathrm{x})^{\mathrm{p}}(1-\mathrm{x})^{\mathrm{q}}$
$\left(1+p x+\frac{p(p-1)}{2!} x^{2}+\ldots\right)$
$\left(1-\mathrm{qx}+\frac{\mathrm{q}(\mathrm{q}-1)}{2!} \mathrm{x}^{2}-\ldots\right)$
$\mathrm{p}-\mathrm{q}=4$
$\frac{\mathrm{p}(\mathrm{p}-1)}{2}+\frac{\mathrm{q}(\mathrm{q}-1)}{2}-\mathrm{pq}=-5$
$\mathrm{p}^{2}+\mathrm{q}^{2}-\mathrm{p}-\mathrm{q}-2 \mathrm{pq}=-10$
$(\mathrm{q}+4)^{2}+\mathrm{q}^{2}-(\mathrm{q}+4)-\mathrm{q}-2(4+\mathrm{q}) \mathrm{q}=-10$
$\mathrm{q}^{2}+8 \mathrm{q}+16-\mathrm{q}^{2}-\mathrm{q}-4-\mathrm{q}-8 \mathrm{q}-2 \mathrm{q}^{2}=-10$
$-2 q=-22$
$\mathrm{q}=11$
$\mathrm{p}=15$
2(15) $+3(11)$
$30+33=63$
2. Let $A=\{2,3,4\}$ and $B=\{8,9,12\}$. Then the number of elements in the relation $R=\left\{\left(\left(a_{1}, b_{1}\right),\left(a_{2}, b_{2}\right)\right) \in(A \times\right.$ $B, A \times B): a_{1}$ divides $b_{2}$ and $a_{2}$ divides $\left.b_{1}\right\}$ is
(1) 18
(2) 24
(3) 12
(4) 36

Sol. (4)

$a_{1}$ divides $\mathrm{b}_{2}$
Each elements has 2 choices
$\Rightarrow 3 \times 2=6$
$\mathrm{a}_{2}$ divides $\mathrm{b}_{1}$
Each elements has 2 choices
$\Rightarrow 3 \times 2=6$
Total $=6 \times 6=36$
3. Let time image of the point $\mathrm{P}(1,2,6)$ in the plane passing through the points $\mathrm{A}(1,2,0), \mathrm{B}(1,4,1)$ and $\mathrm{C}(0,5,1)$ be $\mathrm{Q}(\alpha, \beta, \gamma)$. Then $\left(\alpha^{2}+\beta^{2}+\gamma^{2}\right)$ is equal to
(1) 70
(2) 76
(3) 62
(4) 65

## Sol. (4)

Equation of plane $\mathrm{A}(\mathrm{x}-1)+\mathrm{B}(\mathrm{y}-2)+\mathrm{C}(\mathrm{z}-0)=0$
Put $(1,4,1) \Rightarrow 2 B+C=0$
Put $(0,5,1) \Rightarrow-A+3 B+C=0$
Sub $: \overline{B-A}=0 \Rightarrow A=B, C=-2 B$
$1(x-1)+1(y-2)-2(z-0)=0$
$x+y-2 z-3=0$
Image is $(\alpha, \beta, \gamma) \mathrm{pt} \equiv(1,2,6)$
$\frac{\alpha-1}{1}=\frac{\beta-2}{1}=\frac{\gamma-6}{-2}=\frac{-2(1+2-12-3)}{6}$
$\frac{\alpha-1}{1}=\frac{\beta-2}{1}=\frac{\gamma-6}{-2}=4$
$\alpha=5, \beta=6, \gamma=-2 \Rightarrow \alpha^{2}+\beta^{2}+\gamma^{2}$
$=25+36+4=65$
4. The statement $\sim[p V(\sim(p \wedge q))]$ is equivalent to
(1) $(\sim(p \wedge q)) \wedge q$
$(2) \sim(p \vee q)$
(3) ~ $(\mathrm{p} \wedge \mathrm{q})$
(4) $(p \wedge q) \wedge(\sim p)$

Sol. (4)
$\sim[p v(\sim(p \wedge q))]$
$\sim p \wedge(p \wedge q)$
5. Let $S=\left\{x \in\left(-\frac{\pi}{2}, \frac{\pi}{2}\right): 9^{1-\tan ^{2} x}+9^{\tan ^{2} x}=10\right\}$ and $b=\sum_{x \in S} \tan ^{2}\left(\frac{x}{3}\right)$, then $\frac{1}{6}(\beta-14)^{2}$ is equal to
(1) 16
(2) 32
(3) 8
(4) 64

Sol. (2)
Let $9^{\tan ^{2} x}=P$
$\frac{9}{P}+P=10$
$P^{2}-10 P+9=0$
$(P-9)(P-1)=0$
$P=1,9$
$9^{\tan ^{2} x}=1,9^{\tan ^{2} x}=9$
$\tan ^{2} x=0, \tan ^{2} x=1$
$x=0, \pm \frac{\pi}{4} \quad \therefore x \in\left(-\frac{\pi}{2}, \frac{p}{2}\right)$
$\beta=\tan ^{2}(0)+\tan ^{2}\left(+\frac{\pi}{12}\right)+\tan ^{2}\left(-\frac{\pi}{12}\right)$
$=0+2\left(\tan 15^{\circ}\right)^{2}$
$2(2-\sqrt{3})^{2}$
$2(7-4 \sqrt{3})$
Than $\frac{1}{6}(14-8 \sqrt{3}-14)^{2}=32$
6. If the points P and Q are respectively the circumecenter and the orthocentre of a $\triangle \mathrm{ABC}$, the $\overrightarrow{\mathrm{PA}}+\overrightarrow{\mathrm{PB}}+\overrightarrow{\mathrm{PC}}$ is equal to
(1) $2 \overrightarrow{\mathrm{QP}}$
(2) $\overrightarrow{\mathrm{PQ}}$
(3) $2 \overrightarrow{\mathrm{PQ}}$
(4) $\overrightarrow{P Q}$

## Sol. (4)


$\overrightarrow{\mathrm{PA}}+\overrightarrow{\mathrm{PB}}+\overrightarrow{\mathrm{PC}}=\overrightarrow{\mathrm{a}}+\overrightarrow{\mathrm{b}}+\overrightarrow{\mathrm{c}}$
$\overrightarrow{\mathrm{PG}}=\frac{\overrightarrow{\mathrm{a}}+\overrightarrow{\mathrm{b}}+\overrightarrow{\mathrm{c}}}{3}$
$\Rightarrow \overrightarrow{\mathrm{a}}+\overrightarrow{\mathrm{b}}+\overrightarrow{\mathrm{c}}=3 \overrightarrow{\mathrm{PG}}=\overrightarrow{\mathrm{PQ}}$
Ans. (4)
7. Let $A$ be the point $(1,2)$ and $B$ be any point onthe curve $x^{2}+y^{2}=16$. If the centre of the locus of the point $P$, which divides the line segment $A B$ in the ratio $3: 2$ is the point $C(\alpha, \beta)$ then the length of the line segment AC is
(1) $\frac{6 \sqrt{5}}{5}$
(2) $\frac{2 \sqrt{5}}{5}$
(3) $\frac{3 \sqrt{5}}{5}$
(4) $\frac{4 \sqrt{5}}{5}$

Sol. (3)

$\frac{12 \cos \theta+2}{5}=\mathrm{h} \Rightarrow 12 \cos \theta=5 \mathrm{~h}-2$
sq \& add
$144=(5 h-2)^{2}+(5 k-4)^{2}$
$\left(x-\frac{2}{5}\right)^{2}+\left(y-\frac{4}{5}\right)^{2}=\frac{144}{25}$
Centre $\equiv\left(\frac{2}{5}, \frac{4}{5}\right) \equiv(\alpha, \beta)$
$A C=\sqrt{\left(1-\frac{2}{5}\right)^{2}+\left(2-\frac{4}{5}\right)^{2}}$
$=\sqrt{\frac{9}{25}+\frac{36}{25}}=\frac{\sqrt{45}}{5}=\frac{3 \sqrt{5}}{5}$
8. Let m be the mean and $\sigma$ be the standard deviation of the distribution

| $\mathrm{x}_{\mathrm{i}}$ | 0 | 1 | 2 | 3 | 4 | 5 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $\mathrm{f}_{\mathrm{i}}$ | $\mathrm{k}+2$ | 2 k | $\mathrm{k}^{2}-1$ | $\mathrm{k}^{2}-1$ | $\mathrm{k}^{2}+1$ | $\mathrm{k}-3$ |

where $\sum f_{i}=62$. If $[x]$ denotes the greatest integer $\leq x$, then $\left[\mu 2+\sigma^{2}\right]$ is equal to
(1) 8
(2) 7
(3) 6
(4) 9

Sol. (1)
$\sum f_{i}=62$
$3 k^{2}+16 k-12 k-64=0$
$k=4$ or $-\frac{16}{3}$ (rejected)

$$
\begin{aligned}
& \mu=\frac{\sum f_{i} x_{i}}{\sum f_{i}} \\
& \mu=\frac{8+2(15)+3(15)+4(17)+5}{62}=\frac{156}{62} \\
& \sigma^{2}=\sum f_{i} x_{i}^{2}-\left(\sum f_{i} x_{i}\right)^{2} \\
& =\frac{8 \times 1^{2}+15 \times 13+17 \times 16+25}{62}-\left(\frac{156}{62}\right)^{2} \\
& \sigma^{2}=\frac{500}{62}-\left(\frac{156}{62}\right)^{2} \\
& \sigma^{2}+\mu^{2}=\frac{500}{62} \\
& {\left[\sigma^{2}+\mu^{2}\right]=8}
\end{aligned}
$$

9. If $\mathrm{S}_{\mathrm{n}}=4+11+21+34+50+\ldots$. to n terms, then $\frac{1}{60}\left(\mathrm{~S}_{29}-\mathrm{S}_{9}\right)$ is equal to
(1) 220
(2) 227
(3) 226
(4) 223

Sol. (4)
$S_{n}=4+11+21+34+50+\ldots+n$ terms
Difference are in A.P.
Let $T_{n}=a n^{2}+b n+c$
$T_{1}=a+b+c=4$
$T_{2}=4 a+2 b+c=11$
$T_{3}=9 a+3 b+c=21$
By solving these 3 equations
$a=\frac{3}{2}, b=\frac{5}{2}, c=0$
So $T_{n}=\frac{3}{2} n^{2}+\frac{5}{2} n$
$S_{n}=\Sigma T_{n}$
$=\frac{3}{2} \sum n^{2}+\frac{5}{2} \sum n$
$=\frac{3}{2} \frac{n(n+1)(2 n+1)}{6}+\frac{5}{2} \frac{(n)(n+1)}{2}$
$=\frac{n(n+1)}{4}[2 n+1+5]$
$S_{n}=\frac{n(n+1)}{4}(2 n+6)=\frac{n(n+1)(n+3)}{2}$
$\frac{1}{60}\left(\frac{29 \times 30 \times 32}{2}-\frac{9 \times 10 \times 12}{2}\right)=223$
10. Eight persons are tobe transported from city A to city B in three cars different makes. If each car can accomodate at most three persons, then the number of ways, in which they can be transported, is
(1) 1120
(2) 560
(3) 3360
(4) 1680

Sol. (4)


Ways $=\frac{8!}{3!3!2!2!} \times 3!$
$=\frac{81 \times 7 \times 6 \times 5 \times 4}{4}$
$=56 \times 30$
$=1680$
11. If $A=\frac{1}{5!6!7!}\left[\begin{array}{ccc}5! & 6! & 7! \\ 6! & 7! & 8! \\ 7! & 8! & 9!\end{array}\right]$, then $|\operatorname{adj}(\operatorname{adj}(2 A))|$ is equal to
(1) $2^{16}$
(2) $2^{8}$
(3) $2^{12}$
(4) $2^{20}$

## Sol. (1)

$|A|=\frac{1}{5!6!7!} 5!6!7!\left|\begin{array}{lll}1 & 6 & 42 \\ 1 & 7 & 56 \\ 1 & 8 & 72\end{array}\right|$
$R_{3} \rightarrow R_{3} \rightarrow R_{2}$
$R_{2} \rightarrow R_{2} \rightarrow R_{1}$
$|A|=\left|\begin{array}{lll}1 & 8 & 42 \\ 0 & 1 & 14 \\ 0 & 1 & 16\end{array}\right|=2$
$|\operatorname{adjadj}(2 A)|=|2 A|^{(n-1)^{2}}$
$=|2 A|^{4}$
$=\left(2^{3}|A|\right)^{4}$
$=2^{12}|A|^{4} \Rightarrow 2^{16}$
12. Let the number $(22)^{2022}+(2022)^{22}$ leave the remainder $\alpha$ when divided by 3 and $\beta$ when divided by 7. Then $\left(\alpha^{2}\right.$ $+\beta^{2}$ ) is equal to
(1) 13
(2) 20
(3) 10
(4) 5

Sol. (4)
$(22)^{2022}+(2022)^{22}$
divided byy 3
$(21+1)^{2022}+(2022)^{22}$
$=3 k+1$
( $\alpha=1$ )
Divided by 7

$$
\begin{aligned}
& (21+1)^{2022}+(2023-1)^{22} \\
& 7 k+1+1 \quad(\beta=2) \\
& 7 k+2 \\
& \text { So } \alpha^{2}+\beta^{2} \Rightarrow 5
\end{aligned}
$$

13. Let $g(x)=f(x)+f(1-x)$ and $f^{n}(x)>0, x \in(0,1)$. If $g$ is decreasing in the interval $(0, \alpha)$ and increasing in the interval $(\alpha, 1)$, then $\tan ^{-1}(2 \alpha)+\tan ^{-1}\left(\frac{\alpha+1}{\alpha}\right)$ is equal to
(1) $\frac{5 \pi}{4}$
(2) $\pi$
(3) $\frac{3 \pi}{4}$
(4) $\frac{3 \pi}{2}$

Sol. (2)
$g(x)=f(x)+f(1-x) \& f^{\prime \prime}(x)>0, x \in(0,1)$
$g^{\prime}(x)=f^{\prime}(x)-f^{\prime}(1-x)=0$
$\Rightarrow f^{\prime}(x)=f^{\prime}(1-x)$
$x=1-x$
$x=\frac{1}{2}$
$g^{\prime}(x)=0$
at $x=\frac{1}{2}$
$g^{\prime \prime}(x)=f^{\prime \prime}(x)+f^{\prime \prime}(1-x)>0$
g is concave up
hence $\alpha=\frac{1}{2}$
$\tan ^{-1} 2 \alpha+\tan ^{-1} \frac{1}{\alpha}+\tan ^{-1} \frac{\alpha+1}{\alpha}$
$\tan ^{-1} 1+\tan ^{-1} 2+\tan ^{-1} 3=\pi$
14. For $\alpha, \beta, \gamma, \delta \in N$, if $\int\left(\left(\frac{x}{e}\right)^{2 x}+\left(\frac{e}{x}\right)^{2 x}\right) \log _{e} x d x=\frac{1}{\alpha}\left(\frac{x}{e}\right)^{\beta x}-\frac{1}{\gamma}\left(\frac{e}{x}\right)^{\delta x}+C$, where $e=\sum_{n=0}^{\infty} \frac{1}{n!}$ and $C$ is constant of integration, then $\alpha+2 \beta+3 \gamma-4 \delta$ is equal to
(1) 4
(2) -4
(3) -8
(4) 1

Sol. (1)

$$
x=e^{\ln x}
$$

$\int\left(\left(\frac{x}{e}\right)^{2 x}+\left(\frac{e}{x}\right)^{2 x}\right) \log _{e} x d x=\int\left[e^{2(x \ln x-x)}+e^{-2(x \ln x-x)}\right] \ln x d x$
$x \ln x-x=t$
$\ln x \cdot d x=d t$
$\int\left(e^{2 t}+e^{-2 t}\right) d t$
$\frac{e^{2 t}}{2}-\frac{e^{-2 t}}{2}+C$
$=\frac{1}{2}\left(\frac{x}{e}\right)^{2 x}-\frac{1}{2}\left(\frac{e}{x}\right)^{2 x}+C$
$\alpha=\beta=\gamma=\delta=2$
$\alpha+28+3 \gamma-4 \delta=4$
15. Let f be a continuous function satisying $\int_{0}^{t^{2}}\left(f(x)+x^{2}\right) d x=\frac{4}{3} t^{3}, \forall t>0$. Then $f\left(\frac{\pi^{2}}{4}\right)$ is equal to
(1) $-\pi^{2}\left(1+\frac{\pi^{2}}{16}\right)$
(2) $\pi\left(1-\frac{\pi^{3}}{16}\right)$
(3) $-\pi\left(1+\frac{\pi^{3}}{16}\right)$
(4) $\pi^{2}\left(1-\frac{\pi^{3}}{16}\right)$

Sol. (2)
$\int_{0}^{t^{2}}\left(f(x)+x^{2}\right) d x=\frac{4}{3} t^{3}, \forall t>0$
$\left(f\left(t^{2}\right)+t^{4}\right)=2 t$
$f\left(t^{2}\right)=2 t-t^{4}$
$t=\frac{\pi}{2} \Rightarrow f\left(\frac{\pi^{2}}{4}\right)=\frac{2 \pi}{2}-\frac{\pi^{4}}{16}$
$=\pi-\frac{\pi^{4}}{16}=\pi\left(1-\frac{\pi^{3}}{16}\right)$
16. Let a dic be rolled $n$ times. Let the probability of getting odd numbers seven times be equal to the probability of getting odd numbers nine times. If the probability of getting even numbers twice is $\frac{\mathrm{k}}{2^{15}}$, then k is equal to
(1) 60
(2) 30
(3) 90
(4) 15

Sol. (1)
$P($ odd number7times $)=P($ odd number9times $)$
${ }^{n} C_{7}\left(\frac{1}{2}\right)^{7}\left(\frac{1}{2}\right)^{n-7}={ }^{n} C_{9}\left(\frac{1}{2}\right)^{9}\left(\frac{1}{2}\right)^{n-9}$
${ }^{n} C_{7}={ }^{n} C_{9}$
$\Rightarrow n=16$
Required
$\mathrm{P}={ }^{16} C_{2} \times\left(\frac{1}{2}\right)^{16}$
$=\frac{16 \cdot 15}{2} \times \frac{1}{2^{16}}=\frac{15}{2^{13}}$
$\Rightarrow \frac{60}{2^{15}} \Rightarrow k=60$
17. Let a circle of radius 4 be concentric to the ellipse $15 x^{2}+19 y^{2}=285$. Then the common tangents are inclined to the minor axis ofthe ellipse at the angle.
(1) $\frac{\pi}{6}$
(2) $\frac{\pi}{12}$
(3) $\frac{\pi}{3}$
(4) $\frac{\pi}{4}$

Sol. (3)
$\frac{x^{2}}{19}+\frac{y^{2}}{15}=1$


Let tang be

$$
\begin{aligned}
& y=m x \pm \sqrt{19 m^{2}+15} \\
& m x-y \pm \sqrt{19 m^{2}+15}=0
\end{aligned}
$$

Parallel from $(0,0)=4$
$\left|\frac{ \pm \sqrt{19 m^{2}+15}}{\sqrt{m^{2}+1}}\right|=4$
$19 m^{2}+15=16 m^{2}+16$
$3 m^{2}=1$
$m= \pm \frac{1}{\sqrt{3}}$
$\theta=\frac{\pi}{6}$ with x -axis
Required angle $\frac{\pi}{3}$.
18. Let $\vec{a}=2 \hat{i}+7 \hat{j}-\hat{k}, \vec{b}=3 \hat{i}+5 \hat{k}$ and $\vec{C}=\hat{i}+\hat{j}+2 \hat{k}$, Let $\vec{d}$-be a vector which is perpendicular to both $\vec{a}$, and $\vec{b}$, and $\overrightarrow{\mathrm{c}} \cdot \overrightarrow{\mathrm{d}}=12$. The $(-\hat{\mathrm{i}}+\hat{\mathrm{j}}-\hat{\mathrm{k}}) \cdot(\overrightarrow{\mathrm{c}} \times \overrightarrow{\mathrm{d}})$-is equal to
(1) 24
(2) 42
(3) 48
(4) 44

Sol. (4)

$$
\begin{aligned}
& \vec{a}=2 \hat{i}+7 \hat{j}-\hat{k} \\
& \vec{b}=3 \hat{i}+5 \hat{k} \\
& \vec{c}=\hat{i}-\hat{j}+2 \hat{k}
\end{aligned}
$$

$\vec{d}=\lambda(\vec{a} \times \vec{b})=\lambda\left|\begin{array}{ccc}\hat{i} & \hat{j} & \hat{k} \\ 2 & 7 & -1 \\ 3 & 0 & 5\end{array}\right|$
$\vec{d}=\lambda(35 \hat{i}-13 \hat{j}-21 \hat{k})$
$\lambda(35+13-42)=12$
$\lambda=2$
$\vec{d}=2(35 \hat{i}-13 \hat{j}-21 \hat{k})$
$(\hat{i}+\hat{j}-\hat{k})(\vec{c} \times \vec{d})$
$=\left|\begin{array}{ccc}-1 & 1 & -1 \\ 1 & -1 & 2 \\ 70 & -26 & -42\end{array}\right|=44$
19. Let $S=\left\{z=x+i y: \frac{2 z-3 i}{4 z+2 i}\right.$ is a real number $\}$. Then which of the following is NOT correct ?
(1) $\mathrm{y} \in\left(-\infty,-\frac{1}{2}\right) \bigcup\left(-\frac{1}{2}, \infty\right)$
(2) $(x, y)=\left(0,-\frac{1}{2}\right)$
(3) $x=0$
(4) $y+x^{2}+y^{2} \neq-\frac{1}{4}$

Sol. (2)
$\frac{2 z-3 i}{4 z+2 i} \in R$
$\frac{2(x+i y)-3 i}{4(x+i y)+2 i}=\frac{2 x+(2 y-3) i}{4 x+(4 y+2) i} \times \frac{4 x-(4 y+2) i}{4 x-(4 y+2) i}$
$4 x(2 y-3)-2 x(4 y+2)=0$
$x=0 \quad y \neq-\frac{1}{2}$
Ans. $=2$
20. Let the line $\frac{x}{1}=\frac{6-y}{2}=\frac{z+8}{5}$ intersect the lines $\frac{x-5}{4}=\frac{y-7}{3}=\frac{z+2}{1}$ and $\frac{x+3}{6}=\frac{3-y}{3}=\frac{z-6}{1}$ at the points $A$ and $B$ respectively. Then the distance of the mid-point of the line segment $A B$ from the plane $2 x-2 y+z=$ 14 is
(1) 3
(2) $\frac{10}{3}$
(3) 4
(4) $\frac{11}{3}$

Sol. (3)
$\frac{x}{1}=\frac{y-6}{-2}=\frac{z+8}{5}=\lambda$
$\frac{x-5}{4}=\frac{y-7}{3}=\frac{z+2}{1}=\mu$
$\frac{x+3}{6}=\frac{y-3}{-3}=\frac{z-6}{1}=\gamma$
Intersection of (1) \& (2) "A"
$(\lambda,-2 \lambda+6,5 \lambda-8) \&(4 \mu+5,3 \mu+7, \mu-2)$
$\lambda=1, \mu=-1$
$A(1,4,-3)$
Intersection (1) \& (3) "B"
$(\lambda,-2 \lambda+6,5 \lambda-8) \&(6 \gamma-3,-3 \gamma+3, \gamma+6)$
$\lambda=3$
$\gamma=1$
$B(3,0,7)$
Mod point of A \& B $\Rightarrow(2,2,2)$
Perpendicular distance from the plane
$2 x-2 y+z=14$
$\left|\frac{2(2)-2(2)+2-14}{\sqrt{4+4+1}}\right|=4$

## SECTION - B

21. The sum of all the four-digit numbers that can be formed using all the digits $2,1,2,3$ is equal to $\qquad$ -.
Sol. (26664)
2,1,2,3
$-\quad-\frac{3!}{2!}=3$
$-\quad \underline{2} 3!=6$
$-\quad \underline{3} \quad \frac{3!}{2!}=3$
Sum of digits of unit place $=3 \times 1+6 \times 2+3 \times 3=24$
Required sum
$=24 \times 1000+24 \times 100+24 \times 10+24 \times 1$
$=24 \times 1111$
$=26664$
22. In the figure, $\theta_{1}+\theta_{2}=\frac{\pi}{2}$ and $\sqrt{3}(\mathrm{BE})=4(\mathrm{AB})$. If the area of $\Delta \mathrm{CAB}$ is $2 \sqrt{3}-3$ unit $^{2}$, when $\frac{\theta_{2}}{\theta_{1}}$ is the largest, thenthe perimeter (in unit) of $\triangle$ CED is equal to $\qquad$ .


Sol. (6)

$\sqrt{3} B E=4 \mathrm{AB}$
$\operatorname{Ar}(\triangle \mathrm{CAB})=2 \sqrt{3}-3$
$\frac{1}{2} \mathrm{x}^{2} \tan \theta_{1}=2 \sqrt{3}-3$
$\mathrm{BE}=\mathrm{BD}+\mathrm{DE}$
$=\mathrm{x}\left(\tan \theta_{1}+\tan \theta_{2}\right)$
$\mathrm{BE}=\mathrm{AB}\left(\tan \theta_{1}+\cot \theta_{1}\right)$
$\frac{4}{\sqrt{3}} \tan \theta_{1}+\cot \theta_{1} \Rightarrow \tan \theta_{1}=\sqrt{3}, \frac{1}{\sqrt{3}}$
$\theta_{1}=\frac{\pi}{6} \quad \theta_{2}=\frac{\pi}{3}$
$\theta_{1}=\frac{\pi}{3} \quad \theta_{2}=\frac{\pi}{6}$
as $\frac{\theta_{2}}{\theta_{1}}$ is largest $\therefore \theta_{1}=\frac{\pi}{6} \theta_{2}=\frac{\pi}{3}$
$\therefore \mathrm{x}^{2}=\frac{(2 \sqrt{3}-3) \times 2}{\tan \theta_{1}}=\frac{\sqrt{3}(2-\sqrt{3}) \times 2}{\tan \frac{\pi}{6}}$
$x^{2}=12-6 \sqrt{3}=(3-\sqrt{3})^{2}$
$x=3-\sqrt{3}$
Perimeter of $\triangle C E D$
$=C D+D E+C E$
$=3 \sqrt{3}+(3-\sqrt{3}) \sqrt{3}+(3-\sqrt{3}) \times 2=6$
Ans. (6)
23. Let the tangent at any point $P$ on a curve passing through the points $(1,1)$ and $\left(\frac{1}{10}, 100\right)$, intersect positive $x$ axis and $y$-axis at the points $A$ and $B$ respectively. If $P A: P B=1: k$ and $y=y(x)$ is the solution of the differential equation $\mathrm{e}^{\frac{\mathrm{dy}}{\mathrm{dx}}}=\mathrm{kx}+\frac{\mathrm{k}}{2}, \mathrm{y}(0)=\mathrm{k}$, then 4 y then $4 \mathrm{y}(1)-5 \log \mathrm{e}^{3}$ is equal to $\qquad$ -
Sol. (5)
$Y-y=\frac{d y}{d x}(X-x)$
$Y=0$
$X=\frac{-y d x}{d y}+x$

$\frac{\mathrm{k} \alpha+0}{\mathrm{k}+1}=\mathrm{x}, \alpha=\frac{\mathrm{k}+1}{\mathrm{k}} \mathrm{x}$
$\frac{k+1}{k} x=-y \frac{d x}{d y}+x$
$x+\frac{x}{k}=-y \frac{d x}{d y}+x$
$x \frac{d y}{d x}+k y=0$
$\frac{d y}{d x}+\frac{k}{x} y=0$
$y \cdot x^{\mathrm{k}}=\mathrm{C}$
$\mathrm{C}=1$
$100 \cdot\left(\frac{1}{10}\right)^{k}=1$
$K=2$
$\frac{d y}{d x}=\ln (2 x+1)$
$y=\frac{(2 x+1)}{2}(\ln (2 x+1)-1)+c$
$2=\frac{1}{2}(0-1)+C$
$C=2+\frac{1}{2}=\frac{5}{2}$
$y(1)=\frac{3}{2}(\ell \ln 3-1)+\frac{5}{2}$
$=\frac{3}{2} \ln 3+1$
$4 y(1)=6 \ln 3+4$
$4 y(1)-5 \ln 3=4+\ln 3$
24. Suppose $a_{1}, a_{2}, 2, a_{3}, a_{4}$ be in an arithemetico-geometric progression. If the common ratio of the corresponding geometric progression in 2 and the sum of all 5 terms of the arithmetico-geometric progression is $\frac{49}{2}$, then a4 is equal to $\qquad$ .
Sol. (16)

$$
\begin{aligned}
& \frac{(a-2 d)}{4}, \frac{(a-d)}{2}, a, 2(a+d), 4(a+2 d) \\
& a=2 \\
& \left(\frac{1}{4}+\frac{1}{2}+1+6\right) \times 2+(-1+2+8) d=\frac{49}{2} \\
& 2\left(\frac{3}{4}+7\right)+9 d=\frac{49}{2} \\
& 9 d=\frac{49}{2}-\frac{62}{4}=\frac{98-62}{4}=9 \\
& d=1 \\
& \Rightarrow a_{4}=4(a+2 d) \\
& =16
\end{aligned}
$$

25. If the area of the region $\left\{(x, y):\left|x^{2}-2\right| \leq x\right\}$ is $A$, then $6 A+16 \sqrt{2}$ is equal to

Sol. (27)

$A=\int_{1}^{\sqrt{2}}\left(\mathrm{x}-\left(2-\mathrm{x}^{2}\right)\right) \mathrm{dx}+\int_{\sqrt{2}}^{2}\left(\mathrm{x}-\left(\mathrm{x}^{2}-2\right)\right) \mathrm{dx}$
$=\left(1-2 \sqrt{2}+\frac{2 \sqrt{2}}{3}\right)-\left(\frac{1}{2}-2+\frac{1}{3}\right)+\left(2-\frac{8}{3}+4\right)-\left(1-\frac{2 \sqrt{2}}{3}+2 \sqrt{2}\right)$
$=-4 \sqrt{2}+\frac{4 \sqrt{2}}{3}+\frac{7}{6}+\frac{10}{3}=\frac{-8 \sqrt{2}}{3}+\frac{9}{2}$
$6 \mathrm{~A}=-16 \sqrt{2}+27 \therefore 6 \mathrm{~A}+16 \sqrt{2}=27$
Ans. 27
26. Let the food of perpendicular from the point $A(4,3,1)$ on the plane $P: x-y+2 z+3=0$ be $N$. If $B(5, \alpha, \beta)$, $\alpha, \beta \in \mathrm{Z}$ is a point on plane P such that the area of the triangle ABN is $3 \sqrt{2}$, then $\alpha^{2}+\beta^{2}+\alpha \beta$ is equal to

Sol. (7)

$\mathrm{AN}=\sqrt{6}$
$5-\alpha+2 \beta+3=0$
$\Rightarrow \alpha=8+2 \beta$
N is given by
$\frac{x-4}{1}=\frac{y-3}{-1}=\frac{z-1}{2}=\frac{-(4-3+2+3)}{1+1+4}$
$x=3, y=4, z=-1$
N
$(3,4,-1)$
$\mathrm{BN}=\sqrt{4+(\alpha-4)^{2}+(\beta+1)^{2}}$
$=\sqrt{4+(2 \beta+4)^{2}+(\beta+1)^{2}}$
Area of $\triangle \mathrm{ABN}=\frac{1}{2} \mathrm{AN} \times \mathrm{BN}=3 \sqrt{2}$

$$
\begin{aligned}
& \frac{1}{2} \times \sqrt{6} \times \mathrm{BN}=3 \sqrt{2} \\
& \mathrm{BN}=2 \sqrt{3} \\
& 4+(2 \beta+4)^{2}+(\beta+1)^{2}=12 \\
& (2 \beta+4)^{2}+(\beta+1)^{2}-8=0 \\
& 5 \beta^{2}+18 \beta+9=0 \\
& (5 \beta+3)(\beta+3)=0 \\
& \beta=-3 \\
& \alpha=2 \\
& \alpha^{2}+\beta^{2}+\alpha \beta=9+4-6=7
\end{aligned}
$$

27. Let $S$ be the set of values of $\lambda$, for which the system of equations
$6 \lambda x-3 y+3 z=4 \lambda^{2}$,
$2 x+6 \lambda y+4 z=1$,
$3 x+2 y+3 \lambda z=\lambda$ has no solution. Then $12 \sum_{1 \in S}|\lambda|$ is equal to $\qquad$ .
Sol. (24)
$\Delta=\left|\begin{array}{ccc}6 \lambda & -3 & 3 \\ 2 & 6 \lambda & 4 \\ 3 & 2 & 3 \lambda\end{array}\right|=0$
$2 \lambda\left(9 \lambda^{2}-4\right)+(3 \lambda-6)+(2-9 \lambda)=0$
$18 \lambda^{3}-14 \lambda-4=0$
$(\lambda-1)(3 \lambda+1)(3 \lambda+2)=0$
$\Rightarrow \lambda=1,-1 / 3,-2 / 3$
For each values of $\lambda, \Delta_{1}=\left|\begin{array}{ccc}6 \lambda & -3 & 4 \lambda^{2} \\ 2 & 6 \lambda & 1 \\ 3 & 2 & \lambda\end{array}\right| \neq 0$
$12\left(1+\frac{1}{3}+\frac{2}{3}\right)=24$
28. If the domain of the function $f(x)=\sec ^{-1}\left(\frac{2 x}{5 x+3}\right)$ is $[\alpha, \beta) U(\gamma, \delta]$, then $|3 \alpha+10(\beta+\gamma)+21 \delta|$ is equal to $\qquad$ -.
Sol. (24)

$$
\begin{aligned}
& f(x)=\sec ^{-1} \frac{2 x}{5 x+3} \\
& \left|\frac{2 x}{5 x+3}\right| \\
& \left.\left|\frac{2 x}{5 x+3}\right| \geq 1 \Rightarrow|2 x| \geq 15 x+3 \right\rvert\, \\
& (2 x)^{2}-(5 x+3)^{2} \geq 0 \\
& (7 x+3)(-3 x-3) \geq 0
\end{aligned}
$$


$\therefore$ domain $\left[-1, \frac{-3}{5}\right) \cup\left(\frac{-3}{5}, \frac{-3}{7}\right]$
$\alpha=-1, \beta=\frac{-3}{5}, \gamma=\frac{-3}{5}, \delta=\frac{-3}{7}$
$3 \alpha+10(\beta+\gamma)+21 \delta=-3$
$-3+10\left(\frac{-6}{5}\right)+\left(\frac{-3}{7}\right) 21=-24$
29. Let the quadratic curve passing through the point $(-1,0)$ and touching the line $y=x$ at $(1,1)$ be $y=f(x)$. Then the $x$-intercept of the normal to the curve at the point $(\alpha, \alpha+1)$ in the first quadrant is $\qquad$ _.
Sol. (11)
$\mathrm{f}(\mathrm{x})=(\mathrm{x}+1)(\mathrm{ax}+\mathrm{b})$
$1=2 a+2 b$
$\mathrm{f}^{\prime}(\mathrm{x})=(\mathrm{ax}+\mathrm{b})+\mathrm{a}(\mathrm{x}+1)$
$1=(3 a+b)$
$\Rightarrow \mathrm{b}=1 / 4, \mathrm{a}=1 / 4$
$f(x)=\frac{(x+1)^{2}}{4}$
$\mathrm{f}^{\prime}(\mathrm{x})=\frac{\mathrm{x}}{2}+\frac{1}{2} \quad \alpha+1=\frac{(\alpha+1)^{2}}{4}, \alpha>-1$
$\alpha+1=4$
$\alpha=3$
normal at $(3,4)$
$y-4=-\frac{1}{2}(x-3)$
$y=0 \quad x=8+3$
Ans. 11
30. Let the equations of two adjacent sides of a parallelogram $A B C D$ be $2 x-3 y=-23$ and $5 x+4 y=23$. If the equation of its one diagonal $A C$ is $3 x+7 y=23$ and the distance of $A$ from the other diagonal is $d$, then $50 d^{2}$ is equal to $\qquad$ -.
Sol. (529)


A \& C point will be $(-4,5) \&(3,2)$
mid point of AC will be $\left(-\frac{1}{2}, \frac{7}{2}\right)$
equation of diagonal $B D$ is

$$
\begin{aligned}
& y-\frac{7}{2}=\frac{\frac{7}{2}}{-\frac{1}{2}} \quad\left(\mathrm{x}+\frac{1}{2}\right) \\
& \Rightarrow 7 \mathrm{x}+\mathrm{y}=0
\end{aligned}
$$

Distance of A from diagonal BD
$=d=\frac{23}{\sqrt{50}}$
$\Rightarrow \quad 50 d^{2}=(23)^{2}$
$50 d^{2}=529$

## SECTION - A

31. Given below are two statements:

Statement I : Rotation of the earth shows effect on the value of acceleration due to gravity (g)
Statement II : The effect of rotation of the earth on the value of ' $g$ ' at the equator is minimum and that at the pole is maximum.
In the light of the above statements, choose the correct answer from the options given below.
(1) Both Statement I and Statement II are true
(2) Both Statement I and Statement II are false
(3) Statement I is false but statement II is true
(4) Statement I is true but statement II is false

## Sol. (4)

Due to rotation of earth, $g_{\text {eff }}=g-\omega^{2} R \cos ^{2} \theta$
Where ' $\theta$ ' is angle made with equator
Also, At poles, $\theta=90^{\circ}$
$\Delta g=\omega^{2} R \cos ^{2} \theta$
$=\omega^{2} \mathrm{R} \cos ^{2} 90=0$
[no effect on poles]
$g_{\text {eff }}=g-\omega^{2} R \cos ^{2} \theta$
for equator $\theta=0^{\circ}$
So, $g_{\text {eff }}=g-w^{2} R$
$\& \Delta \mathrm{~g}=\omega^{2} \mathrm{R}$ (Which is maximum change)

32. The ratio of intensities at two points $P$ and $Q$ on the screen in a Young's double slit experiment where phase difference between two waves of same amplitude are $\pi / 3$ and $\pi / 2$, respectively are
(1) $3: 2$
(2) $3: 1$
(3) $2: 3$
(4) $1: 3$

## Sol. (1)

$$
\begin{aligned}
& \mathrm{I}_{\mathrm{res}}=4 \mathrm{I}_{\mathrm{o}} \cos ^{2}\left(\frac{\theta}{2}\right) \\
& \text { If } \begin{aligned}
\theta & =\frac{\pi}{3}, \mathrm{I}_{\mathrm{res}}
\end{aligned}=4 \mathrm{I}_{\mathrm{o}} \cdot \cos ^{2}\left[\frac{\pi}{6}\right] \\
& \\
& \\
& =4 \mathrm{I}_{\mathrm{o}} \cdot\left(\frac{\sqrt{3}}{2}\right)^{2}
\end{aligned}
$$

$$
\mathrm{I}_{1}=\left(4 \mathrm{I}_{\mathrm{o}}\right)\left(\frac{3}{4}\right)=3 \mathrm{I}_{\circ}
$$

$$
\text { If } \theta=\frac{\pi}{2} \quad, I_{\text {res }}=4 I_{o} \cdot \cos ^{2}\left(\frac{\pi}{2}\right)
$$

$$
=4 \mathrm{I}_{\mathrm{o}}\left(\frac{1}{\sqrt{2}}\right)^{2}
$$

$$
=\left(4 \mathrm{I}_{\mathrm{o}}\right)\left(\frac{1}{2}\right)
$$

$\mathrm{I}_{2}=2 \mathrm{I}_{\mathrm{o}}$
$\frac{\mathrm{I}_{1}}{\mathrm{I}_{2}}=\frac{3}{2}$
33. The time period of a satellite, revolving above earth's surface at a height equal to $R$ will be (Given $\mathrm{g}=\pi^{2} \mathrm{~m} / \mathrm{s}^{2}, \mathrm{R}=$ radius of earth)
(1) $\sqrt{32 R}$
(2) $\sqrt{4 R}$
(3) $\sqrt{2 R}$
(4) $\sqrt{8 R}$

Sol. (1)
$\mathrm{T}^{2}=\frac{4 \pi^{2} \mathrm{r}^{3}}{\mathrm{GM}}$
$\mathrm{T}^{2}=\frac{4 \pi^{2}(2 \mathrm{R})^{3}}{\mathrm{GM}}$
$=4 \times 8 \times \frac{\pi 2 \times \mathrm{R}^{3}}{\mathrm{GM}}$
$=4 \times 8 \times \frac{(\mathrm{g})\left(\mathrm{R}^{3}\right)}{\mathrm{GR}^{2}}$


At surface of earth $\Rightarrow g=\frac{G M}{R^{2}}$
So, $\mathrm{GM}=\mathrm{g} . \mathrm{R}^{2}$
Also $\pi^{2}=\mathrm{g}$
$\mathrm{T}^{2}=32 \mathrm{R}$
$\Rightarrow \mathrm{T}=\sqrt{32 \mathrm{R}}$
34. In a metallic conductor, under the effect of applied electric field, the free electrons of the conductor
(1) Move with the uniform velocity throughout from lower potential to higher potential
(2) Move in the curved paths from lower potential to higher potential
(3) Move in the straight line paths in the same direction
(4) Drift from higher potential to lower potential.

Sol. (2)
Electrons moves in curved path because there velocity $\overrightarrow{\mathrm{u}}$ may make any angle $\theta$ with acceleration $\vec{a}$ between time interval of two successive collisions.


Also electron moves from lower potential to higher potential.
35. A message signal of frequency 3 kHz is used to modulate a carrier signal of frequency 1.5 MHz . The bandwidth of the amplitude modulated wave is
(1) 6 kHz
(2) 3 kHz
(3) 6 MHz
(4) 3 MHz

Sol. (1)
Bond width of Amplitude modulated signal $(A M)=2 \times f_{\text {(message signal) }}$

$$
\begin{aligned}
& =2 \times 3 \mathrm{KHz} \\
& =6 \mathrm{KHz}
\end{aligned}
$$

36. In an experiment with vernier calipers of least count 0.1 mm , when two jaws are joined together the zero of vernier scale lies right to the zero of the main scale and $6^{\text {th }}$ division of vernier scale coincides with the main scale division. While measuring the diameter of a spherical bob, the zero of vernier scale lies in between 3.2 cm and 3.3 cm marks, and $4^{\text {th }}$ division of vernier scale coincides with the main scale division. The diameter of bob is measured as
(1) 3.25 cm
(2) 3.22 cm
(3) 3.18 cm
(4) 3.26 cm

## Sol. (3)

The zero error in verniel scale is $=6 \times 0.1 \mathrm{~mm}=0.6 \mathrm{~mm}$ (+ve zero error)
Note: + ve zero error will have to be subtracted
From the reading of the object.
Now, the diameter measured with the help of Vernier scale is
Given by $\rightarrow$ M.S.D + V.S.D $\times$ L.S

$$
\begin{aligned}
& \Rightarrow 3.2 \mathrm{~cm}+0.1 \mathrm{~mm} \times 4 \\
& =3.24 \mathrm{~cm}
\end{aligned}
$$

The actual diameter is $\Rightarrow 3.24 \mathrm{~mm}-$ (zero error) $=3.24-0.6=3.18 \mathrm{~cm}$
37. Two projectiles are projected at $30^{\circ}$ and $60^{\circ}$ with the horizontal the same speed. The ratio of the maximum height attained by the two projectiles respectively is:
(1) $2: \sqrt{3}$
(2) $1: \sqrt{3}$
(3) $\sqrt{3}: 1$
(4) $1: 3$

Sol. (4)
In projectile motion, $\mathrm{H}_{\text {many }}=\frac{\mathrm{u}^{2} \sin ^{2} \theta}{2 \mathrm{~g}}$
at $\theta=30^{\circ}, \quad \mathrm{H}_{1}=\frac{\mathrm{u}^{2} \sin ^{2} 30^{\circ}}{2 \mathrm{~g}}$
at $\theta=60^{\circ}, \quad H_{2}=\frac{u^{2} \sin ^{2} 60^{\circ}}{2 \mathrm{~g}}$
$\frac{H_{1}}{\mathrm{H}_{2}}=\frac{\sin ^{2} 30^{\circ}}{\sin ^{2} 60^{\circ}}=\frac{\left(\frac{1}{2}\right)}{\left(\frac{\sqrt{3}}{2}\right)^{2}}=\frac{1}{3}$
38. Given below are two statements : one is labelled as Assertion $\mathbf{A}$ and then other is labelled as Reason $\mathbf{R}$

Assertion A : An electric fan continues to rotate for some time after the current is switched off.
Reason R : Fan continues to rotate due to inertia of motion.
In the light of above statements, choose the most appropriate answer from the options given below.
(1) A is not correct but $\mathbf{R}$ is correct
(2) Both $\mathbf{A}$ and $\mathbf{R}$ are correct and $\mathbf{R}$ is the correct explanation of $\mathbf{A}$
(3) Both $\mathbf{A}$ and $\mathbf{R}$ are correct but $\mathbf{R}$ is NOT the correct explanation of $\mathbf{A}$
(4) A is correct but $\mathbf{R}$ is not correct

Sol. (2)
Inertia is the property of mass due to which the object continues to move until any external force do not stops it. In the case of rotation of fan, if we switch off then also it moves for some time as air resistance takes time to stop it and due to inertia of fan it moves for some time.
39. The distance between two plates of a capacitor is $d$ and its capacitance is $C_{1}$, when air is the medium between the plates. If a metal sheet of thickness $\frac{2 \mathrm{~d}}{3}$ and of the same area as plate is introduced between the plates, the capacitance of the capacitor becomes $C_{2}$. The ratio $\frac{\mathrm{C}_{2}}{\mathrm{C}_{1}}$ is
(1) $4: 1$
(2) $3: 1$
(3) $2: 1$
(4) $1: 1$

Sol. (2)

$\mathrm{t}=\frac{2 \mathrm{~d}}{3}$
$\mathrm{K}=\infty$ for metals
$\mathrm{C}_{1}=\frac{\varepsilon_{0} \mathrm{~A}}{\mathrm{~d}}$
$\mathrm{C}_{2}=\frac{\varepsilon_{0} \mathrm{~A}}{\mathrm{~d}-\mathrm{t}+\frac{\mathrm{t}}{\mathrm{k}}}=\frac{\varepsilon_{0} \mathrm{~A}}{\mathrm{~d}-\frac{2 \mathrm{~d}}{3}+0}=\frac{3 \varepsilon_{0} \mathrm{~A}}{\mathrm{~d}}$
$=\frac{\mathrm{C}_{2}}{\mathrm{C}_{1}}=\frac{\frac{3 \varepsilon_{0} \mathrm{~A}}{\mathrm{~d}}}{\frac{\varepsilon_{0} \mathrm{~A}}{\mathrm{~d}}}=\frac{3}{1}$
40. The amplitude of magnetic field in an electromagnetic wave propagating along $y$-axis is $6.0 \times 10^{-7} \mathrm{~T}$. The maximum value of electric field in the electromagnetic wave is
(1) $2 \times 10^{15} \mathrm{Vm}^{-1}$
(2) $2 \times 10^{14} \mathrm{Vm}^{-1}$
(3) $6.0 \times 10^{-7} \mathrm{Vm}^{-1}$
(4) $180 \mathrm{Vm}^{-1}$

Sol. (4)
In electromagnetic wave, $\mathrm{E}_{0}=\mathrm{B}_{0} \mathrm{C}$
$\mathrm{E}_{0}=6 \times 10^{-7} \times 3 \times 10^{8}$
$\mathrm{E}_{0} \rightarrow$ Amplitude of electric field
$=18 \times 10^{1}$
$\mathrm{B}_{0} \rightarrow$ Amplitude of magnetic field
$=180 \mathrm{v} / \mathrm{m}$
C $\rightarrow$ Speed of light
41. If each diode has a forward bias resistance of $25 \Omega$ in the below circuit,


Which of the following options is correct:
(1) $\frac{I_{1}}{I_{2}}=2$
(2) $\frac{I_{2}}{I_{3}}=1$
(3) $\frac{I_{3}}{I_{4}}=1$
(4) $\frac{I_{1}}{I_{2}}=1$

Sol. (1)


Here we can see that $D_{1}$ and $D_{3}$ conducts but $D_{2}$ is reversed biased.
Current $I_{1}$ will be equally distributed among $I_{3}$ and $I_{4}$ and $I_{3}=0$
$\mathrm{I}_{1}=\mathrm{I}_{2}+\mathrm{I}_{4}+\mathrm{I}_{3}$
$\mathrm{I}_{1}=2 \mathrm{I}_{2}$
$\frac{\mathrm{I}_{1}}{\mathrm{I}_{2}}=2$
42. A gas mixture consists of 2 moles of oxygen and 4 moles of neon at temperature T. Neglecting all vibrational modes, the total internal energy of the system will be,
(1) 4RT
(2) 11 RT
(3) 8RT
(4) 16 RT

Sol. (2)
Internal energy of $\mathrm{O}_{2}=\frac{5}{2} n \mathrm{nT}=\frac{5}{2} \times 2 \mathrm{RT}=5 \mathrm{RT}$
Internal energy of $\mathrm{Ne}=\mathrm{nRT}=\frac{3}{2} \times \mathrm{nRT}=\frac{3}{2} \times 4 \mathrm{RT}=6 \mathrm{RT}$
Total energy of mixture (system) $=5 R T+6 R T=11 R T$
43. For a periodic motion represented by the equation $y=\sin \omega t+\cos \omega t$ the amplitude of the motion is
(1) 0.5
(2) 1
(3) 2
(4) $\sqrt{2}$

Sol. (4)
If equation of SHM is in the form $\mathrm{y}=\mathrm{a} \sin (\omega \mathrm{t})+\mathrm{B} \cos (\omega \mathrm{t})$
Then its amplitude is $=\sqrt{\mathrm{A}^{2}+\mathrm{B}^{2}}$
Here $\mathrm{A}=\mathrm{B}=1$ in equation $\mathrm{y}=\sin (\omega \mathrm{t})+($ cost $)$
Therefore, Amplitude $=\sqrt{(1)^{2}+(1)^{2}}=\sqrt{2}$
44. A person travels $x$ distance with velocity $v_{1}$ and then $x$ distance with velocity $v_{2}$ in the same direction. The average velocity of the person is $v$, then the relation between $v, v_{1}$ and $v_{2}$ will be.
(1) $v=v_{1}+v_{2}$
(2) $\frac{1}{\mathrm{v}}=\frac{1}{\mathrm{v}_{1}}+\frac{1}{\mathrm{v}_{2}}$
(3) $\frac{2}{\mathrm{v}}=\frac{1}{\mathrm{v}_{1}}+\frac{1}{\mathrm{v}_{2}}$
(4) $v=\frac{v_{1}+v_{2}}{2}$

Sol. (3)


Time taken $b / w$ A \& $B \Rightarrow t_{1}=\frac{x}{v_{1}}$
Time between $b / w B \& C \Rightarrow t_{2}=\frac{x}{v_{2}}$
Average velocity $(v)=\frac{\text { Totaldisplacement }}{\text { Total time }}=\frac{x+x}{t_{1}+t_{2}}=\frac{2 x}{\frac{x}{v_{1}}+\frac{x}{v_{2}}}$
$(v)=\frac{2 \mathrm{v}_{1} \mathrm{v}_{2}}{\mathrm{v}_{1}+\mathrm{v}_{2}} \quad$ or $\quad \frac{2}{\mathrm{v}}=\frac{1}{\mathrm{v}_{1}}+\frac{1}{\mathrm{v}_{2}}$
45. The half life of a radioactive substance is $T$. The time taken, for disintegrating $\frac{7}{8}$ th part of its original mass will be:
(1) T
(2) 2 T
(3) 3 T
(4) 8 T

Sol. (3)
If $\frac{7}{8}$ th is disintegrated it means only $\frac{1}{8}$ th part is radioactive active no. of nuclears after ' $n$ ' half lives
$\Rightarrow \frac{\mathrm{N}_{\mathrm{o}}}{2^{\mathrm{n}}}=\frac{\mathrm{N}_{\mathrm{o}}}{8}$
$2^{\mathrm{n}}=8=\mathrm{n}=3$
So, the elapsed is 3 half lives $=3 \mathrm{~T}$
46. A gas is compressed adiabatically, which one of the following statement is NOT true
(1) There is no change in the internal energy
(2) The temperature of the gas increases.
(3) The change in the internal energy is equal to the work done on the gas
(4) There is no heat supplied to the system

Sol. (1)
In Adiabatic process, $\Delta \mathrm{Q}=0$
If gas is compressed, then w (by gas) $\neq 0$
Therefore by $1^{\text {st }}$ law $\quad \Delta \mathrm{Q}=\Delta \mathrm{u}+\mathrm{w}$

$$
\begin{aligned}
& 0=\Delta u+w \\
& \Delta u=-w \neq 0
\end{aligned}
$$

It implies in adiabatic compression, internal energy of gas changes.
47. Given below are two statements:

Statement I : For diamagnetic substance, $-1 \leq X<0$, where $X$ is the magnetic susceptibility.
Statement II : Diamagnetic substances when placed in an external magnetic field, tend to move from stronger to weaker part of the field.
In the light of the above statements, choose the correct answer from the options given below
(1) Both Statement I and Statement II are false
(2) Statement I is incorrect but Statement II is true
(3) Both Statement I and Statement II are true
(4) Statement I is correct but Statement II is false

Sol. (3)
Diamagnetic substances have the property due to which they tends to move away from stronger magnetic field to weaker magnetic field, as their magnetic susceptibility is negative.
Therefore both statements are correct.
48. Young's moduli of the material of wires $A$ and $B$ are in the ratio of $1: 4$, while its area of cross sections are in the ratio of $1: 3$. If the same amount of load is applied to both the wires, the amount of elongation produced in the wires $A$ and $B$ will be in the ratio of
[Assume length of wires A and B are same]
(1) $12: 1$
(2) $1: 36$
(3) $1: 12$
(4) $36: 1$

Sol. (1)

49. The variation of stopping potential $\left(\mathrm{V}_{0}\right)$ as a function of the frequency $(v)$ of the incident light for a metal is shown in figure. The work function of the surface is

(1) 2.07 eV
(2) 18.6 eV
(3) 2.98 eV
(4) 1.36 eV

Sol. (1)
Work function $(\phi)=h v^{\text {th }}$
$=6.6 \times 10^{-34} \times 5 \times 10^{14}$
$=33 \times 10^{-20}$
$\phi=3.3 \times 10^{-19} \mathrm{~J}$
$=\frac{3.3 \times 10^{-19}}{1.6 \times 10^{-19}} \mathrm{ev} \Rightarrow 2.07$

50. A bar magnet is released from rest along the axis of a very long vertical copper tube. After some time the magnet will
(1) Oscillate inside the tube
(2) Move down with an acceleration greater than $g$
(3) Move down with almost constant speed
(4) Move down with an acceleration equal to $g$

Sol. (3)
According to lenz's law, the rate of charge of flum produced by bar magnet will be approused by the conducting loops.


## SECTION - B

51. If $917 \AA$ be the lowest wavelength of Lyman series then the lowest wavelength of Balmer series will be
$\qquad$ A.

Sol. (3668)


Lowest wavelength of by may sense will be obtained for trasition $n=\infty \longrightarrow n=1$
and for balmer series, Lyman Series $n=\infty \longrightarrow n=2$
for Lyman, $\mathrm{E}_{0}=\frac{\mathrm{hC}}{917 \AA}$
for balmer, $\frac{\mathrm{E}_{0}}{4}=\frac{\mathrm{hC}}{\lambda(\AA)}$
using this
$\lambda=917 \times 4=3668$
52. A square loop of side 2.0 cm is placed inside a long solenoid that has 50 turns per centimeter and carries a sinusoidally varying current of amplitude 2.5 A and angular frequency $700 \mathrm{rad} \mathrm{s}^{-1}$. The central axes of the loop and solenoid coincide. The amplitude of the emf induced in the loop is $x \times 10^{-4} \mathrm{~V}$. The value of x is $\qquad$ .
(Take, $\pi=\frac{22}{7}$ )
Sol. (44)
emf induced in solenoid $\Rightarrow$ BAWN $\sin (\omega \mathrm{t})$, $\mathrm{w}=700 \mathrm{rad} / \mathrm{s}$
Amplitude $\Rightarrow$ BAWN
Area $(\mathrm{A})=2 \mathrm{~cm} \times 2 \mathrm{~cm}$
$=4 \mathrm{~cm}^{2}$
$=4 \times 10^{-4} \mathrm{~m}^{2}$
(B) $)_{\text {solenoid }}=\mu_{0}$ ni
$=4 \pi \times 10^{-7} \times 5000 \times 2.5$
$=5 \pi \times 10^{-3}$

( $\mathrm{N}=1$ )
$\mathrm{n}=\frac{50 \text { turns }}{\mathrm{cm}}=\frac{5000}{\mathrm{~m}}$
$i=2.5$
Amplitude of emf $=\left(5 p \times 10^{-3}\right)\left(4 \times 10^{-4}\right)(700)(1)$
$=5 \times \frac{22}{7} \times 4 \times 700 \times 10^{-7}=44 \times 10^{-4}$
53. A rectangular parallelepiped is measured as $1 \mathrm{~cm} \times 1 \mathrm{~cm} \times 100 \mathrm{~cm}$. If its specific resistance is $3 \times 10^{-7} \Omega \mathrm{~m}$, then the resistance between its tow opposite rectangular faces will be $\qquad$ $\times 10^{-7} \Omega$.

Sol. (3)
$\rho=3 \times 10^{-7} \Omega-\mathrm{cm}$
$R=\rho \cdot \frac{1}{A}$
$=\frac{3 \times 10^{-7} \times\left(10^{-2} \mathrm{~m}\right)}{\left(100 \times 10^{-4} \mathrm{~m}^{2}\right)}=3 \times 10^{-7}$

54. A force of $-P \hat{k}$ acts on the origin of the coordinate system. The torque about the point $(2,-3)$ is $P(a \hat{i}+b \hat{j})$, The ratio of $\frac{a}{b}$ is $\frac{x}{2}$. The value of $x$ is -

Sol. (3)
$\vec{\tau}=\overrightarrow{\mathrm{r}} \times \overrightarrow{\mathrm{F}}$

$\vec{\tau}=$ head - tail
$=(0-2) \hat{i}+(0(-3)) \hat{j}$
$=-2 \hat{i}+3 \hat{j}$
$\tau=(-2 \hat{\mathrm{i}}+3 \hat{\mathrm{j}}) \times(-\mathrm{p} \hat{\mathrm{k}})$
$=-2 p \hat{p}-3 p \hat{i}$
$=-p(3 p \hat{\mathrm{i}}+2 \mathrm{p} \hat{\mathrm{j}}) \frac{\mathrm{a}}{\mathrm{b}}=\frac{3}{2}=\frac{\mathrm{x}}{2} \quad \mathrm{x}=3$
55. A straight wire carrying a current of 14 A is bent into a semicircular arc of radius 2.2 cm as shown in the figure. The magnetic field produced by the current at the centre $(\mathrm{O})$ of the $\operatorname{arc}$. is $\qquad$ $\times 10^{-4} \mathrm{~T}$


Sol. (2)

$\mathrm{B}_{\text {total }}=\mathrm{B}_{\mathrm{I}}+\mathrm{B}_{\text {II }}+\mathrm{B}_{\text {III }}$
$\mathrm{B}_{\mathrm{I}}=0$
$B_{\text {III }}=0$
Because $\overrightarrow{d l} \times \vec{r}=0$
Now, magnetic field due to semicirclaur ring at its center is given by

$$
\begin{aligned}
& \mathrm{B}_{\text {II }}=\frac{\mu_{0} \mathrm{i}}{4 \mathrm{R}} \\
& =\frac{4 \pi \times 10^{-7} \times 14}{4 \times 2.2 \times 10^{-2}} \\
& =\frac{22}{7} \times \frac{10^{-7} \times 14}{22 \times 10^{-3}} \\
& =2 \times 10^{-7} \\
& =2
\end{aligned}
$$

56. Figure below shows a liquid being pushed out of the tube by a piston having area of cross section $2.0 \mathrm{~cm}^{2}$. The area of cross section at the outlet is $10 \mathrm{~mm}^{2}$. If the piston is pushed at a speed of $4 \mathrm{~cm} \mathrm{~s}^{-1}$, the speed of outgoing fluid is $\qquad$ $\mathrm{cm} \mathrm{s}^{-1}$


Sol. (80)
By equation of continuity
$\mathrm{A}_{1} \mathrm{~V}_{1}=\mathrm{A}_{2} \mathrm{~V}_{2}$
$\left(2 \mathrm{~cm}^{2}\right)(4 \mathrm{~cm} / \mathrm{s})=\left(10 \times 10^{-2} \mathrm{~cm}^{2}\right)(\mathrm{v})$
$\frac{8 \mathrm{~cm}^{3}}{\mathrm{~s}}=10^{-1} \mathrm{~cm}^{2}(\mathrm{v})$
$V=80 \mathrm{~cm} / \mathrm{s}$
57. A rectangular block of mass 5 kg attached to a horizontal spiral spring executes simple harmonic motion of amplitude 1 m and time period 3.14 s . The maximum force exerted by spring on block is $\qquad$ N .

## Sol. (20)

When an object executes S.H.M, its morning acceleration is given by $\mathrm{a}_{\max }=\omega^{2} \mathrm{~A}$
Where $\omega=\frac{2 \pi}{\mathrm{~T}}$
Therefore, $\mathrm{a}_{\max }=\frac{4 \pi^{2} \mathrm{~A}}{\mathrm{~T}^{2}}$
(Max force) $\mathrm{F}_{\max }=\operatorname{ma}_{\max }=5 \times \frac{4 \times 3.14 \times 3.14}{3.14 \times 3.14} \times(1)$
$=20 \mathrm{~N}$
58. An electron revolves around an infinite cylindrical wire having uniform linear charge density $2 \times 10^{-8} \mathrm{C} \mathrm{m}^{-1}$ in circular path under the influence of attractive electrostatic field as shown in the figure. The velocity of electron with which it is revolving is $\qquad$ $\times 10^{6} \mathrm{~m} \mathrm{~s}^{-1}$. Given mass of electron $=9 \times 10^{-31} \mathrm{~kg}$


Sol. (8)
In uniform circular motion
$\mathrm{F}_{\mathrm{c}}=\mathrm{ma}_{\mathrm{c}}$
$(\mathrm{q})(\mathrm{E})=\frac{m v^{2}}{\mathrm{r}}$
(e) $\left(\frac{2 \mathrm{k} \lambda}{\mathrm{r}}\right)=\frac{\mathrm{mv}^{2}}{\mathrm{r}}$
$\mathrm{v}^{2}=\frac{(\mathrm{e})(2 \mathrm{k} \lambda)}{\mathrm{m}}=\frac{\left(1.6 \times 10^{-19}\right) \times 2 \times\left(9 \times 10^{9}\right) \times\left(2 \times 10^{-8}\right)}{9 \times 10^{-31}}$
$=1.6 \times 4 \times 10^{13}$
$\mathrm{V}^{2}=16 \times 4 \times 10^{12} \Rightarrow \mathrm{v}=8 \times 10^{6} \mathrm{~m} / \mathrm{s}$
Ans. 8
59. A point object, ' $O$ ' is placed in front of two thin symmetrical coaxial convex lenses $L_{1}$ and $L_{2}$ with focal length 24 cm and 9 cm respectively. The distance between two lenses is 10 cm and the object is placed 6 cm away from lens $L_{1}$ as shown in the figure. The distance between the object and the image formed by the system of two lenses is $\qquad$ cm .


Sol. (34)
Due to lens $\mathrm{L}_{1}$

$\mathrm{u}=-6 \mathrm{~m}$
$\mathrm{f}=+24 \mathrm{~m}$
$\frac{1}{v}-\frac{1}{u}=\frac{1}{f}$
$\frac{1}{v}=\frac{1}{24}-\frac{1}{6}=\frac{1-4}{24} \Rightarrow v=-8 m$

Due to lens $L_{2}$


$$
8 \mathrm{~cm}+10 \mathrm{~cm}=18 \mathrm{~cm}
$$

$\mathrm{U}=-18 \mathrm{~m}$
$\mathrm{F}=+9 \mathrm{~m}$
$\frac{1}{v}-\frac{1}{u}=\frac{1}{f} x$
$\frac{1}{\mathrm{v}}-\frac{1}{9}=\frac{1}{18}$
$\mathrm{V}=\frac{1}{\mathrm{v}}=\frac{2-1}{18}$
$V=18 m$
60. If the maximum load carried by an elevator is 1400 kg ( $600 \mathrm{~kg}-$ Passengers +800 kg - elevator), which is moving up with a uniform speed of $3 \mathrm{~m} \mathrm{~s}^{-1}$ and the frictional force acting on it is 2000 N , then the maximum power used by the motor is $\qquad$ $\mathrm{kW}\left(\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}\right)$

Sol. (48)


Tension in the string $\Rightarrow 16000 \mathrm{~N}$
Maximum power $=(\mathrm{F})(\mathrm{V})$
$=16000 \times 3$
$=48000$
$=48 \mathrm{kw}$
Ans. 48

## SECTION - A

## Solid State Easy

61. The correct relationships between unit cell edge length ' a ' and radius of sphere ' r ' for face-centred and bodycentred cubic structures respectively are:
(1) $2 \sqrt{2} r=a$ and $\sqrt{3} r=4 a$
(2) $\mathrm{r}=2 \sqrt{2} \mathrm{a}$ and $4 \mathrm{r}=\sqrt{3} \mathrm{a}$
(3) $r=2 \sqrt{2} a$ and $\sqrt{3} r=4 a$
(4) $2 \sqrt{2} r=a$ and $4 r=\sqrt{3} \mathrm{a}$

Sol. 4

$$
\begin{array}{ll}
\text { FCC } & \text { BCC } \\
\sqrt{2} \mathrm{a}=4 \mathrm{r} & \sqrt{3} \mathrm{a}=4 \mathrm{r} \\
\mathrm{a}=\frac{4 \mathrm{r}}{\sqrt{2}} \times \frac{\sqrt{2}}{\sqrt{2}} & \\
\mathrm{a}=2 \sqrt{2} \mathrm{r} &
\end{array}
$$

## Chemistry in Everyday life Medium

62. The reaction used for preparation of soap from fat is :
(1) an addition reaction
(2) an oxidation reaction
(3) alkaline hydrolysis reaction
(4) reduction reaction

Sol. 3
The process of making is soap is saponification.
Ester + Base $\longrightarrow$ Alcohol + Soap
In saponification, triglycerides are combine with strong base and form fatty acid so this is alkaline Hydrolysis reaction.

## Mole Easy

63. 

## Match List I with List II

| LIST I |  | LIST II |  |
| :--- | :--- | :--- | :--- |
| A | 16 g of $\mathrm{CH}_{4}(\mathrm{~g})$ | I. | Weight 28 g |
| B | 1 g of $\mathrm{H}_{2}(\mathrm{~g})$ | II | $60.2 \times 10^{23}$ electrons |
| C | 1 mole of $\mathrm{N}_{2}(\mathrm{~g})$ | III | Weight 32 g |
| D | 0.5 mol of $\mathrm{SO}_{2}(\mathrm{~g})$ | IV | Occupies 11.4 L volume at STP |

Choose the correct answer from the options given below:
(1) A-II, B-IV, C-I, D-III
(2) A-II, B-IV, C-III, D-I
(3) A-II, B-III, C-IV, D-I
(4) A-I, B-III, C-II, D-IV

Sol. 1
$16 \mathrm{~g} \mathrm{CH}_{4}=$ mole $=1$
$\mathrm{e}-=60.0 \times 10^{23}$
$19 \mathrm{~Hz}=0.5 \mathrm{~mole}=11.4(\mathrm{~L}) \mathrm{STP}$
$1 \mathrm{~mole}^{\mathrm{N}} 2=2 \mathrm{rg}$
$0.5 \mathrm{~mol} \mathrm{SO}_{2}=$ weights 32 g .

## Periodic Table

## Medium

64. The correct order of metallic character is $=$
(1) $\mathrm{K}>\mathrm{Be}>\mathrm{Ca}$
(2) $\mathrm{Be}>\mathrm{Ca}>\mathrm{K}$
(3) $\mathrm{K}>\mathrm{Ca}>\mathrm{Be}$
(4) $\mathrm{Ca}>\mathrm{K}>\mathrm{Be}$

Sol. 3


Metallic character decreases

## GOC Medium

65. The correct order for acidity of the following hydroxyl compound is :
A. $\mathrm{CH}_{3} \mathrm{OH}$
B. $\left(\mathrm{CH}_{3}\right)_{3} \mathrm{COH}$

D.



Choose the correct answer from the options given below:
(1) $\mathrm{E}>\mathrm{C}>\mathrm{D}>\mathrm{A}>\mathrm{B}$
(2) D $>$ E $>$ C $>$ A $>$ B
(3) E $>$ D $>\mathrm{C}>\mathrm{B}>\mathrm{A}$
(4) $\mathrm{C}>\mathrm{E}>\mathrm{D}>\mathrm{B}>\mathrm{A}$

Sol. 1
Acidity $\propto$ stability of conjugate base
Stability order


Activity $\rightarrow \mathrm{E}>\mathrm{C}>\mathrm{D}>\mathrm{A}>\mathrm{B}$

## Coordination Compound Medium

66. Match List I with List II

| LIST I <br> Complex |  | LIST II <br> Crystal Field splitting energy $\left(\Delta_{0}\right)$ |  |
| :--- | :--- | :--- | :--- |
| A | $\left[\mathrm{Ti}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}$ | I. | -1.2 |
| B | $\left[\mathrm{V}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}$ | II | -0.6 |
| C | $\left[\mathrm{Mn}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$ | III | 0 |
| D | $\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$ | IV | -0.8 |

Choose the correct answer from the options given below:
(1) A-IV, B-I, C-II, D-III
(2) A-IV, B-I, C-III, D-II
(3) A-II, B-IV, C-III, D-I
(4) A-II, B-IV, C-I, D-III

Sol. 1

$$
\begin{array}{cccc}
{\left[\mathrm{Ti}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{+2}} & {\left[\mathrm{~V}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{+2}\left[\mathrm{Mn}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{+3}\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}} \\
\downarrow & & \\
\mathrm{Ti}^{+2} & \mathrm{~V}^{+2} & \mathrm{Mn}^{+3} & \mathrm{Fe}^{+3} \\
\downarrow & \downarrow & \downarrow & \downarrow \\
3 \mathrm{~d}^{2} & 3 \mathrm{~d}^{3} & 3 \mathrm{~d}^{4} & 3 \mathrm{~d}^{5}
\end{array}
$$

CFSE $=-0.4 \times \mathrm{t}_{2 \mathrm{~g}}+0.6 \times \mathrm{eg}+\mathrm{xp}$ $=-0.4 \times 2+0.6 \times 0+\mathrm{xp}$
(A) $\quad=-0.8 \rightarrow \mathrm{Ti}^{+2}$
(B) $\quad \mathrm{V}^{+2} \rightarrow 3 \mathrm{~d}^{3}$

$$
\begin{aligned}
\text { CFSE } & =-0.4 \times \mathrm{t}_{2 \mathrm{~g}}+0.6 \times \mathrm{eg}+\mathrm{xp} \\
& =-0.4 \times 3+0.6 \times 0+\mathrm{xp} \\
& =-1.2
\end{aligned}
$$

(C)


$$
\begin{aligned}
\text { CFSE }= & -0.4 \times \mathrm{t}_{2 \mathrm{~g}}+0.6 \times \mathrm{eg}+\mathrm{xp} \\
& -0.4 \times 3+0.6 \times 1+\mathrm{xp} \\
= & -1.2+0.6=0.6
\end{aligned}
$$

(D)


$$
\begin{aligned}
\text { CFSE } & =-0.4 \times \mathrm{t}_{2 \mathrm{~g}}+0.6 \times \mathrm{eg}+\mathrm{xp} \\
& =-0.4 \times 3+0.6 \times 2 \\
& =-1.2+1.2 \\
& =0
\end{aligned}
$$

## Qualitative analysis

Medium
67. In Carius tube, an organic compound ' X ' is treated with sodium peroxide to form a mineral acid ' Y '.The solution of $\mathrm{BaCl}_{2}$ is added to ' Y ' to form a precipitate ' Z '.' Z ' is used for the quantitative estimation of an extra element. ' X ' could be
(1) Chloroxylenol
(2) Methionine
(3) A nucleotide
(4) Cytosine

Sol. 2
Carious method is used for quantitative analysis of sulfur




So Methionine is correct answer

## S-block Medium

68. Number of water molecules in washing soda and soda ash respectively are:
(1) 1 and 0
(2) 1 and 10
(3) 10 and 0
(4) 10 and 1

Sol. 3
Washing Soda $\rightarrow \mathrm{Na}_{2} \mathrm{CO}_{3} \cdot \underline{10} \mathrm{H}_{2} \mathrm{O}$
0.2

Soda Ash $\rightarrow \mathrm{Na}_{2} \mathrm{CO}_{3}$
No. of water $=10+0=(10)$

## Metallurgy

## Medium

69. Gibbs energy vs T plot for the formation of oxides is given below.


For the given diagram, the correct statement is -
(1) At $600{ }^{\circ} \mathrm{C}, \mathrm{C}$ can reduce ZnO
(2) At $600{ }^{\circ} \mathrm{C}, \mathrm{C}$ can reduce FeO
(3) At $600^{\circ} \mathrm{C}$, CO cannot reduce FeO
(4) At $600{ }^{\circ} \mathrm{C}$, CO can reduce ZnO

Sol. 2
$\mathrm{FeO}+\mathrm{C} \longrightarrow \mathrm{Fe}+\mathrm{CO}_{2}$
At $600^{\circ} \mathrm{C} \Delta \mathrm{G}$ of Reaction is -Ve
70. Buna-S can be represented as:
(1)

(2)

(3)


Sol.


## Hydrogen

## Medium

71. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason $\mathbf{R}$

Assertion A : Physical properties of isotopes of hydrogen are different.
Reason : Mass difference between isotopes of hydrogen is very large.
In the light of the above statements, choose the correct answer from the options given below:
(1) Both $\mathbf{A}$ and $\mathbf{R}$ are true but $\mathbf{R}$ is NOT the correct explanation of $\mathbf{A}$
(2) A is false but $\mathbf{R}$ is true
(3) $A$ is true but $\mathbf{R}$ is false
(4) Both $\mathbf{A}$ and $\mathbf{R}$ are true and $\mathbf{R}$ is the correct explanation of $\mathbf{A}$

Sol. Correct - (4)
The Physical properties of isotope of Hydrogen are different due to Large mass difference

## Coordination Compound Medium

72. The correct order of the number of unpaired electrons in the given complexes is
A. $\left[\mathrm{Fe}(\mathrm{CN})_{6}\right]^{3-}$
B. $\left[\mathrm{FeF}_{6}\right]^{3-}$
C. $\left[\mathrm{CoFF}_{6}\right]^{3-}$
D. $\left[\mathrm{Cr}(\text { oxalate })_{3}\right]^{3-}$
E. $\left[\mathrm{Ni}(\mathrm{CO})_{4}\right]$

Choose the correct answer from the options given below:
(1) E $<$ A $<$ D $<$ C $<$ B
(2) A $<$ E $<$ C $<$ B $<$ D
(3) A $<$ E $<$ D $<$ C $<$ B
(4) E $<$ A $<$ B $<$ D $<$ C

Sol. 1


## Topic: GOC

## Medium

73. The decreasing order of hydride affinity for following carbonations is:
A.

B.

C.

D.


Choose the correct answer from the options given below:
(1) C, A, D, B
(2) A, C, B , D
(3) A, C, D, B
(4) C, A, B, D

Sol. 4
Stability of carbocation $\propto \frac{1}{\text { Hydride affinity }}$


## Chapter: carbonyl

## Level : Med.

74. Incorrect method of preparation for alcohols from the following is:
(1) Ozonolysis of alkene.
(2) Hydroboration-oxidation of alkene.
(3) Reaction of alkyl halide with aqueous NaOH .
(4) Reaction of Ketone with RMgBr followed by hydrolysis.

Sol. 1

1) Ozonolysis of alkene-

$$
\xrightarrow{-}=\mathrm{C}^{\prime} \backslash \frac{\mathrm{O}_{3}}{\mathrm{Zn}, \mathrm{H}_{2} \mathrm{O}}-\underset{\mathrm{O}}{\mathrm{C}}-,-\underset{\mathrm{O}}{\mathrm{C}}-
$$

2) Hydroboration - oxidation of alkene

3) $\mathrm{R}-\mathrm{X}+\mathrm{NaOH} \longrightarrow \mathrm{R}-\mathrm{OH}+\mathrm{NaX}$


## \{Chap - Aldehyele, ketone, SO - Med \}

75. In the reaction given below:


The product ' X ' is:
(1)

(2)

(3)

(4)


Sol. 4

s-block Medium
76. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason $\mathbf{R}$

Assertion A : The energy required to form $\mathrm{Mg}^{2+}$ from Mg is much higher than that required to produce $\mathrm{Mg}^{+}$
Reason R: $\mathrm{Mg}^{2+}$ is small ion and carry more charge than $\mathrm{Mg}^{+}$
In the light of the above statements, choose the correct answer from the options given below.
(1) Both $\mathbf{A}$ and $\mathbf{R}$ are true and $\mathbf{R}$ is the correct explanation of $\mathbf{A}$
(2) $\mathbf{A}$ is true but $\mathbf{R}$ is false
(3) $\mathbf{A}$ is false but $\mathbf{R}$ is true
(4) Both $\mathbf{A}$ and $\mathbf{R}$ are true but $\mathbf{R}$ is NOT the correct explanation of $\mathbf{A}$

Sol. Correct - (1)


In formation of $\mathrm{Mg}^{2+} \mathrm{IE}_{1}+\mathrm{IE}_{2}$ is required while in formation of $\mathrm{Mg}^{+} \mathrm{IE}_{1}$ is required
(R) $\quad \mathrm{Mg}^{2+}$ is small ion and carry more change than $\mathrm{Mg}^{\oplus}$
77. The major product ' P ' formed in the given reaction is:

(1)

(2)

(3)

(4)


Sol. 1

78. Ferric chloride is applied to stop bleeding because -
(1) Blood absorbs $\mathrm{FeCl}_{3}$ and forms a complex.
(2) $\mathrm{FeCl}_{3}$ reacts with the constituents of blood which is a positively charged sol.
(3) $\mathrm{Fe}^{3+}$ ions coagulate blood which is a negatively charged sol.
(4) $\mathrm{Cl}^{-}$ions cause coagulation of blood.

Sol. 3
$\mathrm{Fe}^{3+}$ coagulation negatively charged sol blood.

## Environmental Chemistry

## Easy

79. The delicate balance of $\mathrm{CO}_{2}$ and $\mathrm{O}_{2}$ is NOT disturbed by
(1) Burning of Coal
(2) Deforestation
(3) Burning of petroleum
(4) Respiration

Sol. Correct - (4)
The balance of carbon dioxide and oxygen in atmosphere is mainly maintained by the oxygen released and carbon dioxide consumed during photosynthesis by plants.
80. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason $\mathbf{R}$

Assertion A : 3.1500 g of hydrated oxalic acid dissolved in water to make 250.0 mL solution will result in 0.1 M oxalic acid solution.
Reason R : Molar mass of hydrated oxalic acid is $126 \mathrm{~g} \mathrm{~mol}^{-1}$
In the light of the above statements, choose the correct answer from the options given below:
(1) $\mathbf{A}$ is false but $\mathbf{R}$ is true
(2) $\mathbf{A}$ is true but $\mathbf{R}$ is false
(3) Both $\mathbf{A}$ and $\mathbf{R}$ are true but $\mathbf{R}$ is NOT the correct explanation of $\mathbf{A}$
(4) Both $\mathbf{A}$ and $\mathbf{R}$ are true and $\mathbf{R}$ is the correct explanation of $\mathbf{A}$

Sol. 4
Assertion is correct.
$\mathrm{H}_{2} \mathrm{C}_{2} \mathrm{O}_{4} \cdot 2 \mathrm{H}_{2} \mathrm{O}$
$\mathrm{M}=\frac{3.15 \times 1000}{126 \times 250}$
$=\frac{12.6}{126}=0.1$
Reason is correct. It is used as a fact in explanation of assertion.

## SECTION - B

## Chemical bonding

## Medium

81. The number of molecules from the following which contain only two lone pair of electrons is $\qquad$ $\mathrm{H}_{2} \mathrm{O}, \mathrm{N}_{2}, \mathrm{CO}, \mathrm{XeF}_{4}, \mathrm{NH}_{3}, \mathrm{NO}, \mathrm{CO}_{2}, \mathrm{~F}_{2}$
Sol. 4

82. The specific conductance of 0.0025 M acetic acid is $5 \times 10^{-5} \mathrm{~S} \mathrm{~cm}^{-1}$ at a certain temperature. The dissociation constant of acetic acid is $\qquad$ $\times 10^{-7}$. (Nearest integer)
Consider limiting molar conductivity of $\mathrm{CH}_{3} \mathrm{COOH}$ as $400 \mathrm{~S} \mathrm{~cm}{ }^{2} \mathrm{~mol}^{-1}$.
Sol. 66
$\Lambda_{\mathrm{m}}=\frac{\mathrm{k}}{\mathrm{C}} \times 1000$
Given $\mathrm{k}=5 \times 10^{-5} \mathrm{~S} \mathrm{~cm}^{-1}$
$\mathrm{C}=0.0025 \mathrm{M}$
$\Lambda_{\mathrm{m}}=\frac{5 \times 10^{-5} \times 10^{3}}{0.0025}=\frac{5 \times 10^{-2}}{2.5 \times 10^{-3}}$
$=20 \mathrm{Scm}^{2} \mathrm{~mol}^{-1}$
$\alpha=\frac{20}{400}=\frac{1}{20}$
$\mathrm{K}_{\mathrm{a}}=\frac{\mathrm{C} \alpha^{2}}{1-\alpha}=\frac{0.0025 \times \frac{1}{20} \times \frac{1}{20}}{\frac{19}{20}}$
$=\frac{0.0025}{19 \times 20}=6.6 \times 10^{-6}$
$=66 \times 10^{-7}$
83. An aqueous solution of volume $300 \mathrm{~cm}^{3}$ contains 0.63 g of protein. The osmotic pressure of the solution at 300

K is 1.29 mbar . The molar mass of the protein is $\qquad$ $\mathrm{g} \mathrm{mol}^{-1}$
Given : $\mathrm{R}=0.083 \mathrm{~L}^{\text {bar }} \mathrm{K}^{-1} \mathrm{~mol}^{-1}$
Sol. 40535
$\because \pi=$ CRT
$\pi=\frac{\mathrm{n}}{\mathrm{V}} \mathrm{RT}$
$\pi=\frac{\omega}{\mathrm{V}} \frac{\mathrm{RT}}{\mathrm{M}}$
$\mathrm{M}=\frac{\omega \mathrm{RT}}{\pi \times \mathrm{V}}$
$\mathrm{M}=\frac{0.63 \times 0.083 \times 300}{1.29 \times 10^{-3} \times 300 \times 10^{-3}}$
$\mathrm{M}=40535 \mathrm{gm} / \mathrm{moL}$

## p-block Medium

84. The difference in the oxidation state of Xe between the oxidised product of Xe formed on complete hydrolysis of $\mathrm{XeF}_{4}$ and $\mathrm{XeF}_{4}$ is $\qquad$
Sol. 2

$$
\stackrel{+4}{\mathrm{XeF}_{4}}+\mathrm{H}_{2} \mathrm{O} \longrightarrow \mathrm{Xe}+\stackrel{+6}{\mathrm{XeO}_{3}}+\mathrm{O}_{2}+\mathrm{HF}
$$

Difference $=6-4=(2)$
85. The number of endothermic process/es from the following is
A. $\mathrm{I}_{2}(\mathrm{~g}) \rightarrow 2 \mathrm{I}(\mathrm{g})$
B. $\mathrm{HCl}(\mathrm{g}) \rightarrow \mathrm{H}(\mathrm{g})+\mathrm{Cl}(\mathrm{g})$
C. $\mathrm{H}_{2} \mathrm{O}(\mathrm{l}) \rightarrow \mathrm{H}_{2} \mathrm{O}(\mathrm{g})$
D. $\mathrm{C}(\mathrm{s})+\mathrm{O}_{2}(\mathrm{~g}) \rightarrow \mathrm{CO}_{2}(\mathrm{~g})$
E. Dissolution of ammonium chloride in water

Sol. 4
A $\rightarrow$ Endothermic (Atomisation) $\quad \mathrm{B} \rightarrow$ Endothermic (Atomisation)
$\mathrm{C} \rightarrow$ Endothermic (Vapourisation) $\quad \mathrm{D} \rightarrow$ Exothermic (Combustion)
$\mathrm{E} \rightarrow$ Endothermic (Dissolution)
86. The number of incorrect statement/s from the following is
A. The successive half lives of zero order reactions decreases with time.
B. A substance appearing as reactant in the chemical equation may not affect the rate of reaction
C. Order and molecularity of a chemical reaction can be a fractional number
D. The rate constant units of zero and second order reaction are mol L ${ }^{-1} \mathrm{~s}^{-1}$ and $\mathrm{mol}^{-1} \mathrm{Ls}^{-1}$ respectively.

Sol. 1
(A) For zero order $\mathrm{t}_{1 / 2}=\frac{[\mathrm{A}]_{0}}{2 \mathrm{~K}}$ as concentration decreases half life decreases (Correct statement)
(B) If order w.r.t. that reactant is zero then it will not affect rate of reaction. (Correct statement)
(C) Order can be fractional but molecularity can not be (Incorrect statement)
(D) For zero order reaction unit is $\mathrm{mol}^{-\mathrm{Ls}^{-1}}$ and for second order reaction unit is $\mathrm{mol}^{-1} \mathrm{Ls}^{-1}$ (Correct statement)

87.


The electron in the $n$th orbit of $\mathrm{Li}^{2+}$ is excited to $(\mathrm{n}+1)$ orbit using the radiation of energy $1.47 \times 10^{-17} \mathrm{~J}$ (as shown in the diagram). The value of n is $\qquad$
Given: $\mathrm{R}_{\mathrm{H}}=2.18 \times 10^{-18} \mathrm{~J}$
Sol. 1
$\Delta \mathrm{E}=\mathrm{R}_{\mathrm{H}} \mathrm{Z}^{2}\left(\frac{1}{\mathrm{n}_{1}^{2}}-\frac{1}{\mathrm{n}_{2}^{2}}\right)$
$1.47 \times 10^{-17}=2.18 \times 10^{-18} \times 9\left(\frac{1}{\mathrm{n}^{2}}-\frac{1}{(\mathrm{n}+1)^{2}}\right)$
$\frac{1.47}{1.96}=\frac{3}{4}=\frac{1}{\mathrm{n}^{2}}-\frac{1}{(\mathrm{n}+1)^{2}}$
So, $\mathrm{n}=1$

## d-block Medium

88. For a metal ion, the calculated magnetic moment is 4.90BM. This metal ion has $\qquad$ number of unpaired electrons.
Sol. 4
$\mu=4.90 \mathrm{BM}$.
$\mu=\sqrt{\mathrm{n}(\mathrm{n}+2)}$
So, $n=4$
89. In alkaline medium, the reduction of permanganate anion involves a gain of - electrons.

Sol. 3

(3)
90. $\quad \mathrm{A}(\mathrm{g}) \rightleftharpoons 2 \mathrm{~B}(\mathrm{~g})+\mathrm{C}(\mathrm{g})$

For the given reaction, if the initial pressure is 450 mmHg and the pressure at time t is 720 mmHg at a constant temperature $T$ and constant volume $V$. The fraction of $\mathrm{A}(\mathrm{g})$ decomposed under these conditions is $\mathrm{x} \times 10^{-1}$. The value of $x$ is $\qquad$ (nearest integer)
Sol. 3
$\mathrm{A}(\mathrm{g}) \rightleftharpoons 2 \mathrm{~B}(\mathrm{~g})+\mathrm{C}(\mathrm{g})$
$\mathrm{t}=0 \quad 450$
time t $450-x \quad 2 x \quad x$
$\mathrm{P}_{\mathrm{T}}=\mathrm{P}_{\mathrm{A}}+\mathrm{P}_{\mathrm{B}}+\mathrm{P}_{\mathrm{C}}$
$720=450-\mathrm{x}+2 \mathrm{x}+\mathrm{x}$
$2 \mathrm{x}=270$
$\mathrm{x}=135$
Fraction of A decomposed $=\frac{135}{450}=0.3=3 \times 10^{-1}$
So, $x=3$

